

STOICHIOMETRY: THE NUTS AND BOLTS OF CHEMICAL REACTIONS

Stoichiometry is the term that describes mathematical relationships represented by balanced chemical equations. Being able to predict amount of product that should form in a reaction or the amount of reactant needed is a fundamental skill required of a chemist. Relating symbols used to write a chemical equation to physical amounts of reactant and product that can be weighed in a laboratory is the role of stoichiometry.

Nuts and Bolts $\frac{2N}{1BN_2}$ or $\frac{1BN_2}{2N}$ $\frac{2N}{1B}$ or $\frac{1B}{2N}$ $\frac{1B}{1BN_2}$ or $\frac{1BN_2}{1B}$

Obtain a cup with some nuts and bolts. Your goal is to put two nuts on each bolt and make as many of these products as possible.

- Using N for nut and B for bolt, write out an equation analogous to a chemical equation to show the formation of your product. Be sure to use subscripts and coefficients.



- How many nuts did you have in the cup?

7

- How many bolts?

7

- Which reactant (N or B) was limiting? How did you know?

Bolts because I can't make anymore products without them.

- How many of which reactant did you have left over?

Nuts: 3

Bolts: 0

- Weigh out one of the bolts and assume that all of your bolts have the same mass. What is the mass of a single bolt?

7.50g

Nuts And Bolts And Stoichiometry Answers

Scott C. Dulebohn



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