

Name: _____ Date: _____ Class period: _____

Unit: Nuclear Chemistry Test Review



What to study?

- Quizzes: #1
- Homework: worksheets #1-#4
- Notes: Properties of Radiation → Applications of Nuclear Chemistry
- Bell Ringers: Matching, Half-lives, Writing Transmutations

What will the questions be like? What can I expect?

This test will be just like the previous quizzes you have taken.

- Matching
- Multiple choice
- Short answer
- Calculations

Sample questions/Important Topics

1. Fill in the best answer for each of the following:

- a) The transformation of one atom into another atom is called _____.
- b) Three types of radiation include _____, _____, and _____.
- c) During radioactive decay, the _____ will emit particles of radiation in order to become more stable.
- d) The reactions inside the sun is an example of a _____ reaction.
- e) If you wanted to block _____ radiation, paper or clothing can be used.
- f) Radioactive isotopes, or _____, have different mass numbers but the same atomic number and will decay over time.
- g) _____ is used to calculate the date of a specimen that is embedded deep within the earth.
- h) _____ is the time it takes for half a sample of nuclei to decay into more stable nuclei.
- i) Fusion reactions release _____ energy than fission reactions.
- j) For elements that have an atomic number less than 20, stable nuclei have a neutron:proton ratio equal to _____.
- k) _____ is the spontaneous emission of radiation by an unstable nucleus.
- l) A common isotope used to diagnose thyroid disease is called _____.
- m) During _____ decay, the atomic number will increase by 1 and the mass number will remain unchanged.
- n) _____ used alpha particles in his Gold Foil Experiment.

Nuclear Chemistry Unit Review Answers

Chao Zhang



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Principles Of Nuclear Chemistry Peter A C Mcpherson,2016-12-21 Principles of Nuclear Chemistry is an introductory text in nuclear chemistry and radiochemistry aimed at undergraduates with little or no knowledge of physics It covers the key aspects of modern nuclear chemistry and includes worked solutions to end of chapter questions The text begins with basic theories in contemporary physics and uses these to introduce some fundamental mathematical techniques It relates nuclear phenomena to key divisions of chemistry such as atomic structure spectroscopy equilibria and kinetics It also gives an introduction to f block chemistry and the nuclear power industry This book is essential reading for those taking a first course in nuclear chemistry and is a useful companion to other volumes in physical and analytical chemistry It will also be of use to those new to working in nuclear chemistry or radiochemistry Let's Review Albert S. Tarendash,1998

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