

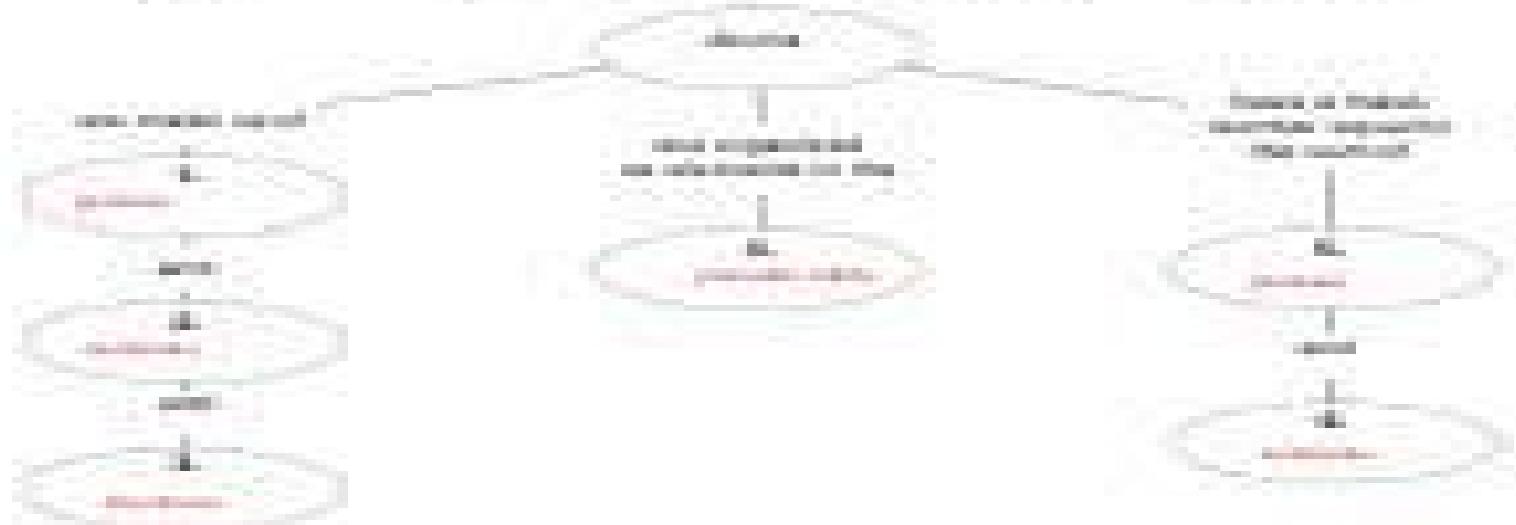
Characteristics Properties of Atoms and the Periodic Table

periodic

give similar traits

vertical rows

horizontal rows



Properties of atoms can be grouped into two main categories:

- (1) Qualitative properties that are independent of type (universal)
• The most important universal trend of atomic properties is that atomic size decreases from left to right.
- (2) Quantitative properties that vary with the number of protons and electrons (periodic).
• In the periodic table, elements are arranged by increasing atomic proton number.
• Electronegativity also increases across the period and decreases down the group because the ionization energy decreases.

Vocabulary Review For Understanding The Atoms Guide

George Graybill

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