

- Q3) a)** Answer any two of the following : [6]
- Define standard solution. What are the characteristics of the standard solutions?
  - What do you mean by significant figure? Give various rules for computations.
  - How is aldehyde detected? Give two characteristic tests for aldehyde.
- b)** Solve any one of the following : [4]

- Following weights were obtained by a person on repeating the same experiment four times i.e. 32.6 mg, 28.3 mg, 31.4 mg and 30.9 mg. Calculate the mean deviation, standard deviation and Relative mean deviation.
- Three grams of solute are dissolved in 50 ml aqueous solution. Calculate the amount of solute extracted after a single extraction with 50 ml of benzene. Distribution ratio for the extraction is 10.

- Q4) a)** What is a titration curve? Discuss the titration curve between  $\text{Fe}^{2+}$  and  $\text{Ce}^{4+}$  ions. [6]

OR

Explain sodium fusion test for detection of nitrogen, sulphur and halogens.

- b)** Answer any one of the following : [4]
- Describe the Mohr's method for estimation of chloride in a given sample.
  - Explain the theory of sulphide and hydroxide precipitation in inorganic qualitative analysis.



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### **Table of Contents Past Question 2013 2014 For Cit701noun**

1. Understanding the eBook Past Question 2013 2014 For Cit701noun
  - The Rise of Digital Reading Past Question 2013 2014 For Cit701noun
  - Advantages of eBooks Over Traditional Books
2. Identifying Past Question 2013 2014 For Cit701noun
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an eBook Past Question 2013 2014 For Cit701noun
  - User-Friendly Interface
4. Exploring eBook Recommendations from Past Question 2013 2014 For Cit701noun
  - Personalized Recommendations
  - Past Question 2013 2014 For Cit701noun User Reviews and Ratings
  - Past Question 2013 2014 For Cit701noun and Bestseller Lists

5. Accessing Past Question 2013 2014 For Cit701noun Free and Paid eBooks
  - Past Question 2013 2014 For Cit701noun Public Domain eBooks
  - Past Question 2013 2014 For Cit701noun eBook Subscription Services
  - Past Question 2013 2014 For Cit701noun Budget-Friendly Options
6. Navigating Past Question 2013 2014 For Cit701noun eBook Formats
  - ePub, PDF, MOBI, and More
  - Past Question 2013 2014 For Cit701noun Compatibility with Devices
  - Past Question 2013 2014 For Cit701noun Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Past Question 2013 2014 For Cit701noun
  - Highlighting and Note-Taking Past Question 2013 2014 For Cit701noun
  - Interactive Elements Past Question 2013 2014 For Cit701noun
8. Staying Engaged with Past Question 2013 2014 For Cit701noun
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Past Question 2013 2014 For Cit701noun
9. Balancing eBooks and Physical Books Past Question 2013 2014 For Cit701noun
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Past Question 2013 2014 For Cit701noun
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Past Question 2013 2014 For Cit701noun
  - Setting Reading Goals Past Question 2013 2014 For Cit701noun
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Past Question 2013 2014 For Cit701noun
  - Fact-Checking eBook Content of Past Question 2013 2014 For Cit701noun
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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