

2-9: Counting Molecules

1. Start *Virtual ChemLab*, select *Reactions and Stoichiometry*, and then select *Counting Molecules (2)* from the list of assignments. The lab will open in the Calorimetry laboratory.
2. Enter the stockroom by clicking inside the *Stockroom* window. Once inside the stockroom, click on the bottle containing table sugar (sucrose, $C_{12}H_{22}O_{11}$) located on the Organics shelf and drag it to the stockroom counter. You can also double click on the bottle to move it to the counter, while the left and right blue arrow keys can be used to see additional bottles. Click on the *Return to Lab* arrow to return to the laboratory.
3. Drag the bottle to the spotlight near the balance, and click on the *Balance* area to zoom in. Drag a piece of weighing paper to the balance pan and then *Tare* the balance so the balance reads 0.0000 g. Click on the bottle lid (*Remove Lid*) to remove the lid.
4. Pick up the *Scoop* and scoop out some sample by first dragging the scoop to the mouth of the bottle and then pulling the scoop down the face of the bottle. As the scoop is dragged down the face of the bottle it will pickup different quantities of solid. Select the largest sample possible and drag the scoop to the weighing paper on the balance until it snaps in place and then let go. This will put approximately 1 g of sample on the balance. Record the mass of the sample in the data and results table on the following page.
5. Repeat steps 2-4 for NH_4Cl (ammonium chloride) located on the Salts shelf and record the mass in the Data and Results Table.
6. Calculate the moles of $C_{12}H_{22}O_{11}$ contained in the first sample and record your results in the data and results table.
7. Calculate the moles of each element in $C_{12}H_{22}O_{11}$ and record your results in the data and results table.

Virtual Chemlab Workbook Answers

Kingsley Augustine

A red circular graphic with a gradient, appearing as a partial circle or a thick curved line, located to the right of the author's name.

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Woodfield, 2004 **Virtual Chemlab** Brian F., Woodfield, Matthew C., Asplund, Steven Haderlie, 2007 Prepared by Brian F Woodfield of Brigham Young University This AIE combines the full student *Virtual ChemLab* manual with answers and more

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