

atoms of solid behaves as a diffraction grating.

$$eV = h\nu$$

$$eV = \frac{hc}{\lambda}$$

$$\lambda = \frac{hc}{eV}$$

$V = 10 \text{ kV}$

$$\lambda = \frac{6.6 \times 10^{-34} \times 3 \times 10^8}{1.6 \times 10^{-19} \times 10 \times 1000}$$

$$\lambda = 1.24 \text{ \AA} \text{ wavelength of X-Ray}$$

Crystal  $\text{NaCl} = (1.8 - 2) \text{ \AA}$

Visible as UV  $\rightarrow$  Reflection, Refraction  
 because their  $\lambda$  very high  
 as compare to interatomic spacing  
 of Ray. its wavelength is very small  
 as compare to interatomic  
 spacing.

## Bragg's Law

X-Ray



Diffraction

Assumptions

- i) Crystal contain various parallel planes



interplaner distance

Crystal  
 Regular  
 arrangements  
 of an  
 atoms

one layer of atom  
 $\rightarrow$  we consider plane

# Solid Physics Students Manual Solution Kittle

**EW Minium**



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