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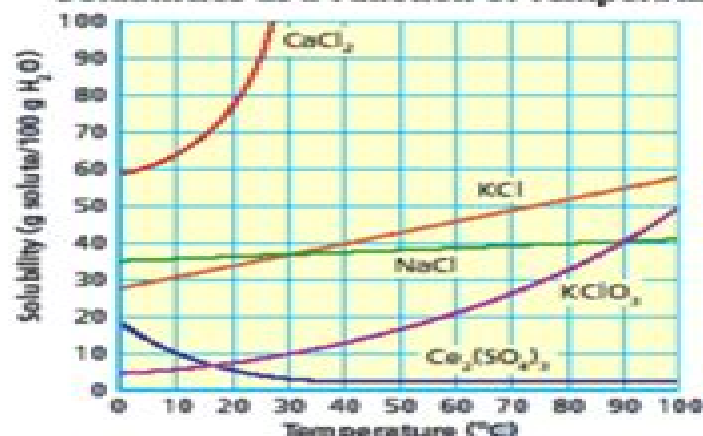
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Solubility Curve Worksheet 2



Directions: Use the graph below to answer the following questions.

Solubilities as a Function of Temperature



Types of Solutions

- > Points that collectively make up the solubility curve (points **ON** the curve) represent _____ solutions.
- > Points that are **BELOW** the curve represent _____ solutions.
- > Points **ABOVE** the solubility curve represent _____ solutions, and the **difference** between the point above the curve and on the curve represents the amount of solute which will precipitate out.
- > Say if the resulting solutions would be saturated, supersaturated or unsaturated.
 - o 60 g of KCl at 70°C _____
 - o 10 g of KClO₃ at 60°C _____
 - o 70 g of CaCl₂ at 20°C _____

1. a. What is the solubility of calcium chloride (CaCl₂) at 5°C? _____
 b. What is the solubility of calcium chloride at 25°C? _____
 c. What is happening to the solubility and temperature for CaCl₂? Solubility is _____ as temperature is _____.
2. a. What is the solubility of cerium sulfate - Ce₂(SO₄)₃ at 10°C? _____
 b. What is the solubility of cerium sulfate at 50°C? _____
 c. What is happening to the solubility and temperature for Ce₂(SO₄)₃? Solubility is _____ as temperature is _____.
3. Which substance on the graph is **least** soluble at 10°C? _____
4. Which substance on the graph shows the **least** change in solubility from 0°C to 100°C? _____
5. Which solution is more concentrated:
 - a) At 10°C, a saturated solution of NaCl or a saturated solution of CaCl₂ (circle one)
 - b) At 55°C, a saturated solution of KClO₃ or a saturated solution of Ce₂(SO₄)₃ (circle one)
6. What is the mass of cerium sulfate that will dissolve in 50 g of water at 10 degrees Celsius?
7. What is the mass of potassium chloride that will dissolve in 250 g of water at 70 degrees Celsius?
8. At 90 degrees Celsius, 10 g of potassium chlorate is dissolved in 100 g of water. Is this solution saturated, unsaturated, or supersaturated? How do you know?
9. A saturated solution of potassium chlorate is dissolved in 100g of water. If the saturated solution is cooled from 90 degrees Celsius to 60 degrees Celsius, how many grams would crystallize out?

Solubility Curve Practice Problems 1 Key

Kevin T. Patton, Victoria Schauf



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Science Test Practice, Grade 8 Spectrum,2012-09-01 Spectrum Science Test Practice provides the most comprehensive strategies for effective science test preparation Each book features engaging and comprehensive science content including physical science earth and space science and life science The lessons perfect for students in grade 8 are presented through a variety of formats and each book includes suggestions for parents and teachers as well as answer keys a posttest and a standards chart Today more than ever students need to be equipped with the essential skills they need for school achievement and for success on proficiency tests The Spectrum series has been designed to prepare students with these skills and to enhance student achievement Developed by experts in the field of education each title in the Spectrum workbook series offers grade appropriate instruction and reinforcement in an effective sequence for learning success Perfect for use at home or in school and a favorite of parents homeschoolers and teachers worldwide Spectrum is the learning partner students need for complete achievement

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Science I Essential Interactions ,2000-10

Analytical Chemistry Bryan M. Ham,Aihui

MaHam,2015-10-26 A comprehensive study of analytical chemistry providing the basics of analytical chemistry and introductions to the laboratory Covers the basics of a chemistry lab including lab safety glassware and common instrumentation Covers fundamentals of analytical techniques such as wet chemistry instrumental analyses spectroscopy chromatography FTIR NMR XRF XRD HPLC GC MS Capillary Electrophoresis and proteomics Includes ChemTech an interactive program that contains lesson exercises useful calculators and an interactive periodic table Details Laboratory Information Management System a program used to log in samples input data search samples approve samples and print reports and certificates of analysis **Holt Chemistry** Ralph Thomas Myers,2004 **A Problem-Solving Approach to Aquatic Chemistry** James N. Jensen,2023-01-05 A Problem Solving Approach to Aquatic Chemistry Enables civil and environmental engineers to understand the theory and application of aquatic equilibrium chemistry The second edition of A Problem Solving Approach to Aquatic Chemistry provides a detailed introduction to aquatic equilibrium chemistry calculation methods for systems at equilibrium applications of aquatic chemistry and chemical kinetics The text directly addresses two required ABET program outcomes in environmental engineering chemistry including stoichiometry equilibrium and kinetics and material and energy balances fate and transport of substances in and between air water and soil phases The book is very student centered with each chapter beginning with an introduction and ending with a summary that reviews the chapter s main points To aid in reader comprehension important terms are defined in context and key ideas are summarized Many thought provoking discussion questions worked examples and end of chapter problems are also included Each part of the text begins with a case study a portion of which is addressed in each subsequent chapter illustrating the principles of that chapter In addition each chapter has an Historical Note exploring connections with the people and cultures connected to topics in the text A Problem Solving Approach to Aquatic Chemistry includes Fundamental concepts such as concentration units thermodynamic basis of equilibrium and manipulating equilibria Solutions of chemical equilibrium problems including setting up the problems and algebraic graphical and computer solution techniques Acid base equilibria including the concepts of acids and bases titrations and alkalinity and acidity Complexation including metals ligands equilibrium calculations with complexes and applications of complexation chemistry Oxidation reduction equilibria including equilibrium calculations graphical approaches and applications Gas liquid and solid liquid equilibrium with expanded coverage of the effects of global climate change Other topics including chemical kinetics of aquatic systems surface chemistry and integrative case studies For advanced senior undergraduates and first year graduate students in environmental engineering courses A Problem Solving Approach to Aquatic Chemistry serves as an invaluable learning resource on the topic with a variety of helpful learning elements included throughout to ensure information retention and the ability to apply covered concepts in practical settings *General Chemistry* Ralph H. Petrucci,1989 **Solvent Systems and Their Selection in Pharmaceuticals and Biopharmaceutics** Patrick Augustijns,Marcus Brewster,2007-08-06 Solvent systems are integral to drug development and

pharmaceutical technology This single topic encompasses numerous allied subjects running the gamut from recrystallization solvents to biorelevant media The goal of this contribution to the AAPS Biotechnology Pharmaceutical Aspects series is to generate both a practical handbook as well as a reference allowing the reader to make effective decisions concerning the use of solvents and solvent systems To this end the monograph was created by inviting recognized experts from a number of fields to author relevant sections Specifically 15 chapters have been designed covering the theoretical background of solubility the effect of ionic equilibria and pH on solubilization the use of solvents to effect drug substance crystallization and polymorph selection the use of solvent systems in high throughput screening and early discovery solvent use in preformulation the use of solvents in bio relevant dissolution and permeation experiments solvents and their use as toxicology vehicles solubilizing media and excipients in oral and parenteral formulation development specialized vehicles for protein formulation and solvent systems for topical and pulmonary drug administration The chapters are organized such that useful decision trees are included together with the scientific underpinning for their application In addition trends in the use of solvent systems and a balance of current views make this monograph useful to both the novice and experienced researcher and to scientists at all developmental stages from early discovery to late pharmaceutical operations

Journal of Research of the National Bureau of Standards United States. National Bureau of Standards,1971 *Journal of Research of the National Bureau of Standards* ,1972

Crystallization of Organic Compounds Hsien-Hsin Tung,Edward L. Paul,Michael Midler,James A. McCauley,2023-07-05 Crystallization of Organic Compounds Practical resource covering applications of crystallization principles with methodologies case studies and numerous industrial examples for emphasis Based on the authors hands on experiences as process engineers through the use of case studies and examples of crystallization processes ranging from laboratory development through manufacturing scale up Crystallization of Organic Compounds guides readers through the practical applications of crystallization and emphasizes strategies that have proven to be successful enabling readers to avoid common pitfalls that can render standard procedures unsuccessful Most chapters feature multiple examples that guide readers step by step through the crystallization of active pharmaceutical ingredients APIs including an analysis of the major methods of carrying out crystallization operations their strengths and potential issues as well as numerous examples of crystallization processes from development through manufacturing scale Advancements in the field of crystallization have been integrated throughout the book in the newly revised Second Edition to ensure the content adequately reflects current state of the art industrial know hows and practice The new edition also adds chapters addressing downstream operations after the crystallization including filtration washing and drying together with industrial use cases Crystallization of Organic Compounds includes detailed information on Solubility and solid behavior covering phase rule polymorph salt co crystal chiral resolution and in silico solubility prediction and kinetics covering seed supersaturation nucleation crystal growth and model based experimental design Critical issues in the crystallization practice covering oiling

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Scientifica Pupil Book 7 Essentials (Level 3-6) David Sang, Lawrie Ryan, Peter Ellis, Derek McMonagle, Jane Taylor, 2004 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities **High School Chemistry Unlocked** The Princeton Review, 2016-11-29 UNLOCK THE SECRETS OF CHEMISTRY with THE PRINCETON REVIEW High School Chemistry Unlocked focuses on giving you a wide range of key lessons to help increase your understanding of chemistry With this book you ll move from foundational concepts to complicated real world applications building confidence as your skills improve End of chapter drills will help test your comprehension of each facet of chemistry from atoms to alpha radiation Don t feel locked out Everything You Need to Know About Chemistry Complex concepts explained in straightforward ways Walk throughs of sample problems for all topics Clear goals and self assessments to help you pinpoint areas for further review Guided examples of how to solve problems for common subjects Practice Your Way to Excellence 165 hands on practice questions seeded throughout the chapters and online Complete answer explanations to boost understanding Bonus online questions similar to those you ll find on the AP Chemistry Exam and the SAT Chemistry Subject Test High School Chemistry Unlocked covers Building blocks of matter Physical behavior of matter Chemical bonding Chemical reactions Stoichiometry Solutions Acids and bases Equilibrium Organic chemistry Radioactivity and more *Chemistry Extension File* Eileen Ramsden, 2002 This chemistry extension file includes teaching notes guidance on coursework activities and equipment It has at least one assignment for each topic in the textbooks suitable for classwork and homework A comprehensive range of practical activities are included It contains extensive Key Skills and ICT materials An exam file resource containing a complete set of exam style questions in a format that can be used throughout Years 10 and 11 or as a resource for a revision programme is included **Enhanced Oil Recovery** Association de recherche sur les techniques d'exploitation du pétrole, 1982

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