

Chem 546 – Lab 1: Solubility of Organic Compounds Answer Key

Pre-Lab Assignment

The following should be completed in your laboratory notebook *prior* to the start of lab.

(1 point) Descriptive Title of the Experiment

Determining the Solubility of Organic Compounds

(1 point) Objectives

The objectives of this experiment are to determine the solubility of organic compounds in a variety of different organic and aqueous solvents.

(3 points) Structures

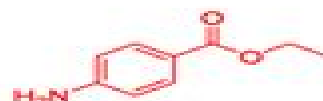
Draw the **structures** (not the molecular formulas) of the compounds listed in the table of reagents.



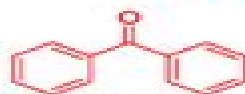
Biphenyl



Piperidine



Ethyl-4-aminobenzoate



Benzophenone



Malonic Acid



Butanol

(6 points) Table of Reagents

Practice identifying functional groups and calculating MW by filling in the table below. Some of these compounds have more than one functional group and some functional groups are present in more than one of the compounds. Refer to the inside back cover of your CH 545 text for a list of common functional groups. *Note that the functional group consisting of a 6-member ring with three carbon-carbon double bonds is not listed there, and you will need to look elsewhere in your CH 545 textbook.*

Substance	MW (g/mol)	Functional Group(s)
Biphenyl	154	Aromatic Ring
Piperidine	85	Amine
Ethyl-4-aminobenzoate	165	Amine (or aniline), ester, aromatic ring
Benzophenone	182	Ketone, aromatic ring
Malonic Acid	104	Carboxylic Acid
Butanol	74	Alcohol

Solubility Lab Answer Key

Wenbin Ji

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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Transitions 2023-2024 Daniel B. Friedman, Katie Hopkins, Kristy Sokol, 2024-03-27 A publication of University 101 Programs University of South Carolina Transitions is the customized textbook for students in the University of South Carolina's University 101 first year seminar. It includes both general and institution specific information for first year students. Topics include time management, academic success strategies, career development, information literacy, health and wellness, and values and identity. An ideal model for institutions working to design a custom published first year seminar text. Teaching Undergraduate Science Linda C. Hodges, 2023-07-03 This book is written for all science or engineering faculty who have ever found themselves baffled and frustrated by their undergraduate students' lack of engagement and learning. The author, an experienced scientist, faculty member, and educational consultant, addresses these issues with the knowledge of faculty interests, constraints, and day-to-day concerns in mind. Drawing from the research on learning, she offers faculty new ways to think about the struggles their science students face. She then provides a range of evidence-based teaching strategies that can make the time faculty spend in the classroom more productive and satisfying. Linda Hodges reviews the various learning problems endemic to teaching science, explains why they are so common and persistent, and presents a digest of key ideas and strategies to address them based on the research she has undertaken into the literature on the cognitive sciences and education. Recognizing that faculty have different views about teaching, different comfort levels with alternative teaching approaches, and are often pressed for time, Linda Hodges takes these constraints into account by first offering a framework for thinking purposefully about course design and teaching choices, and then providing a range of strategies to address very specific teaching barriers, whether it be students' motivation, engagement in class, ability to problem solve, their reading comprehension, or laboratory research or writing skills. Except for the first and last chapters, the other chapters in this book stand on their own; i.e., they can be read in any order and address a specific challenge students have in learning and doing science. Each chapter summarizes the research explaining why students struggle and concludes by offering several teaching options categorized by how easy or difficult they are to implement. Some, for example, can work in a large lecture class without a great expenditure of time; others may require more preparation and a more adventurous approach to teaching. Each strategy is accompanied by a table categorizing its likely impact, how much time it will take in class or out, and how difficult it will be to implement. Like scientific research, teaching works best when faculty start with a goal in mind, plan an approach, building on the literature, use well-tested methodologies, and analyze results for future trials. Linda Hodges' message is that with such intentional thought and a bit of effort, faculty can succeed in helping many more students gain exciting new skills and abilities, whether those students are potential scientists or physicians or entrepreneurs. Her book serves as a mini compendium of current research, as well as a protocol manual, a readily accessible guide to the literature, the best practices known to date, and a framework for thinking about teaching. **Report** United States. National Bureau of Standards, 1968 **Practical**

Chemistry for CSEC Norman Lambert, 1987-03-30 Practical Chemistry is a unique practice book for CXC It provides a wealth of revision exercises and a guide to all the detailed experimental work covered in the CXC Chemistry syllabus Section A Practical guidance for teachers and classes perform **NBS Special Publication** ,1968 *Scientifica* David Sang, Peter Ellis, Derek McMonagle, 2004 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities **Chemistry (Student)** Dennis Englin, 2018-02-08 What is chemistry It is the study of the composition structure and properties of matter It is through an understanding of chemistry that the products that have benefited society were discovered and technologies to sustain the environment were put in place Knowledge taught in this course of how matter changes will give us an insight into the origin of life so we can realize that life could only have been formed by a supernatural act of creation not by a process of change over time High school science course with lab curriculum Lab experiments are included with step by step images for guidance Based on the principle that those who can understand and apply information do much better than those who simply memorize material This course has been taught by Dr Englin for many years with students going on to medical and graduate school He wanted to develop a series of courses that would give students the tools to help them succeed in higher education The comprehensive material has God the Creator as its foundation A teacher guide is available for Chemistry providing this full year science course with a detailed schedule worksheets and tests Current Hydraulic Laboratory Research in the United States ,1970 *Hydraulic Research in the United States* 1970 United States. National Bureau of Standards, 1971 **Proceedings of the Workshop on Geochemical Modeling** ,1986 *Circular* United States. National Bureau of Standards, **Organophosphates Chemistry, Fate, and Effects** Janice E. Chambers, Patricia E. Levi, 2013-10-22 This volume pulls together a wealth of up to date information on the toxicology of this diverse and ubiquitous class of insecticides Leading experts review the reactivity of organophosphorus compounds with cholinesterase as well as their metabolism and biological effects on humans and other nontarget organisms The book not only covers the anticholinesterase actions of organophosphates but also other presumably independent effects such as teratogenicity delayed neuropathy immunotoxicity and behavioral toxicity Offers crucial overviews of chemical and biochemical reactivity and biological responsiveness of mainly nontarget organisms Reviews new developments in assessment of metabolism and disposition of organophosphorus compounds Examines organophosphorus compound induced toxicity mediated by mechanisms other than inhibition of acetylcholinesterase Covers new research on differences in toxicity induced by the variety of organophosphorus Analyzes current controversies on the relevance of certain biochemical parameters in actual organophosphate toxicity in vivo *Circular of the Bureau of Standards* ,1927 **Science Educator's Guide to Laboratory Assessment** Rodney L. Doran, 2002 The book opens with an up to date discussion of assessment theory research and uses Then comes a wealth of sample assessment activities in biology chemistry physics and Earth science Keyed to the

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