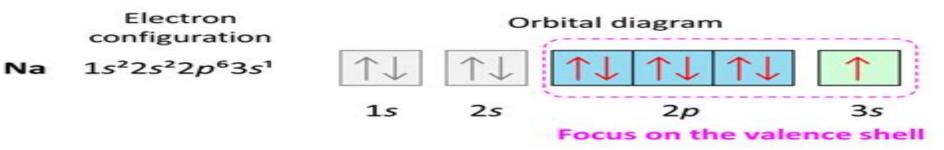
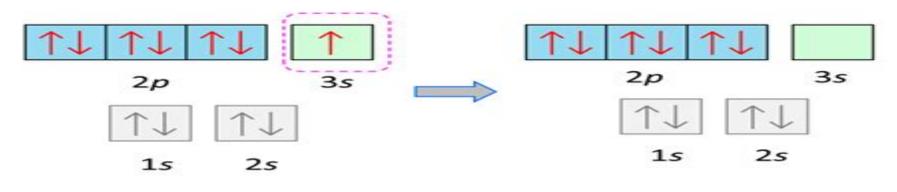
Electron Configuration of Main Group Cations



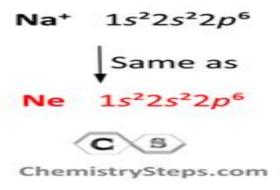
The charge is +1, therefore, one electron is lost:

The electron must be from the outermost valence orbital $-3s^1$



Na $1s^2 2s^2 2p^6 3s^1$

Cations of main group metals attain the electron configuration of the noble gas that precedes them in the periodic table. They are *isoelectronic* with the previous noble gas.



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Nuclear Science Abstracts .1971 Chemistry Eugene LeMay, Jr., Herbert Beall, Karen M. Robblee, Douglas C. ERDA Energy Research Abstracts United States. Energy Research and Brower, Prentice-Hall Staff, 2002-02 Scientific and Technical Aerospace Reports, 1988 **Congressional Record** United Development Administration, 1976 States. Congress, 1978 **Energy Research Abstracts** ,1991-10 Fusion Energy Update ,1978 Government-wide Index to Federal Research & Development Reports ,1967 ERDA Energy Research Abstracts United States. Energy Research and Development Administration. Technical Information Center, 1976 **Plasma Physics and** Physics Briefs ,1994 **Magnetohydrodynamics** Defense Documentation Center (U.S.),1962 U.S. Government Research Reports ,1963 *Using high energy density plasmas for nuclear experiments relevant to nuclear astrophysics* Maria Gatu Johnson, Michael Wiescher, Mark Paris, Alex Zylstra, Gerald Hale, 2023-04-14 U.S. Government Research & Development Reports ,1967 ERDA Energy Research Abstracts United States. Energy Research and Development Administration.1976-11 Government Reports Announcements & Index ,1993 **Encyclopaedia Britannica** Harry S. Ashmore, 1962 **INIS Atomindex** ,1987 1992 Fall Meeting ,1992 Government Reports Announcements, 1973

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