

SMALL-SCALE LAB: Making a Solution**LABORATORY RECORD SHEET**

Use with SECTION B3.2

SAFETY

Wear safety glasses and follow the standard safety precautions outlined on page 7 of this manual.

PURPOSE

To make a solution, and use carefully measured data to calculate the solution's concentration in various units.

MATERIALS

- solid NaCl
- water
- 50-mL plastic volumetric bottle
- balance

PROCEDURE

Weigh a clean, dry, plastic volumetric bottle. Add solid NaCl to approximately the 5-mL mark on the bottle. Weigh the bottle again. Half fill the bottle with water and shake it gently until all the NaCl dissolves. Fill the bottle with water to the 50-mL mark and weigh it again.

ANALYSIS

Using your experimental data, record the answers to the following questions.

1. Percent by mass tells how many grams of solute are present in 100 g of solution.

$$\% \text{ by mass} = \frac{\text{mass of solute}}{\text{mass of solute} + \text{solvent}} \times 100$$

- a. Calculate the mass of the solute (NaCl).
- b. Calculate the mass of the solvent (water).
- c. Calculate the percent by mass of NaCl in the solution.

Small Scale Lab Use With Section 9 2 Answers

Reva Arnez Curry, Betty Bates Tempkin



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Beall, Karen M. Robblee, Douglas C. Brower, Prentice-Hall Staff, 2002-02 *Exploring Biology in the Laboratory: Core Concepts* Murray P. Pendarvis, John L. Crawley, 2019-02-01 *Exploring Biology in the Laboratory Core Concepts* is a comprehensive manual appropriate for introductory biology lab courses This edition is designed for courses populated by nonmajors or for majors courses where abbreviated coverage is desired Based on the two semester version of *Exploring Biology in the Laboratory 3e* this *Core Concepts* edition features a streamlined set of clearly written activities with abbreviated coverage of the biodiversity of life These exercises emphasize the unity of all living things and the evolutionary forces that have resulted in and continue to act on the diversity that we see around us today **Commerce Business Daily**, 2001 **Organic Chemistry Education Research into Practice** Jay Wackerly, Sarah Zingales, Michael Wentzel, Gautam Bhattacharyya, Brett McCollum, 2025-03-25 This Research Topic has three main goals 1 provide a platform for instructors of organic chemistry to showcase evidence based methods and educational theories they have utilized in their classrooms 2 build new and strengthen existing connections between educational researchers and practitioners and 3 highlight how people have used chemical education based research in their teaching practice There are places in the literature dedicated for chemical education research CER however there is not a clear avenue for those that have changed their teaching methods based on published CER and report their experiences Creating this article collection will foster collaboration between chemical education researchers and teachers of organic chemistry This opportunity allows these instructors to share evidence based practices experiences challenges and innovative approaches from CER literature and beyond This Research Topic bridges discipline based education research and the scholarship of teaching and learning which will help advance organic chemistry education and improve student outcomes *Bulletin* United States. Bureau of Mines, **Proceedings of the Public Health Conference on Records and Statistics** Public Health Conference on Records and Statistics, 1995 A planned study program of the Public Health Service administered by the National Center for Health Statistics **Research Reporting Series**, 1975

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