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web determination of a rate law and activation energy in this experiment glass beakers pipettes and a hotplate were utilized to set up the procedures a solution of deionized water a buffer potassium iodide ki sodium thiosulfate na<sub>2</sub>s<sub>2</sub>o<sub>3</sub> starch and hydrogen peroxide h<sub>2</sub>o<sub>2</sub> were used in order to calculate the rate of reaction

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web core concepts some reactions will go fast and some will go slow the speed of the reaction is it s reaction rate which is dictated by a rate law in this article we will learn about reaction rates rate laws the rate constant and the reaction order

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