

9. This question tests all ions and complexes.

You are provided with two boiling tubes containing solutions of the same ionic compound. The compound contains one cation and one anion from the list below.

- (1) cations: Na^+ , NH_4^+ , Ba^{2+}
(2) anions: Cl^- , NO_3^- , SO_4^{2-}

They are tested in ways that you could apply to any of the samples to identify the ions. Compounds that easily dissolve produce no tests and should be omitted from the report too.

The results are:

- (1) colour change of reagents, observed observations and equations
(2) colourless precipitate observed after the ions are identified

You may include observations in tables in your answer.

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10. You are supplied with a sample of ammonium carbonate.

Describe simple tests that would identify the cation and anion present in ammonium carbonate.

Include reagents, expected observations and relevant equations.

Ocr Cmay 201mark Scheme

Colin Shephard,Rosemary Rees



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