

### Separating Mixtures

The components of a mixture may be separated based on the physical properties of \_\_\_\_\_.

\_\_\_\_\_ (the instrument that uses this property to separate mixtures where tiny solid particulates dissolved in a liquid is called a \_\_\_\_\_).

Some techniques allow separation of components of a mixture into \_\_\_\_\_ based on separate liquids and solids.

\_\_\_\_\_ used for separate solutions.

#### Observation

#### Explanation

1. \_\_\_\_\_ (\_\_\_\_\_ is \_\_\_\_\_).

\_\_\_\_\_

2. \_\_\_\_\_ (\_\_\_\_\_ is \_\_\_\_\_).

\_\_\_\_\_

3. \_\_\_\_\_ (\_\_\_\_\_ is \_\_\_\_\_).

\_\_\_\_\_

\_\_\_\_\_

Summary \_\_\_\_\_ and \_\_\_\_\_.

Equation: \_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_ for density: \_\_\_\_\_.

Example problem: Suppose two substances are mixed in a ratio of \_\_\_\_\_ parts to \_\_\_\_\_ parts. If the density of the mixture is \_\_\_\_\_, what is the density of the \_\_\_\_\_?

\_\_\_\_\_

\_\_\_\_\_ is density of mixture

\_\_\_\_\_

# Separating Mixtures Note Taking Guide Answers

**Carol Chapman, Moira Sheehan**



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