

GEO-SCI 103: Spring 2017

Study Questions and Key Topics for Exam 2

Lecture 5: Water and Salt (pp. 108-115)

1. What are the unique properties of water (pp. 110)?

- 1) Water has a unique response of density to temperature D
- 2) Water has high boiling and freezing points F
- 3) Water has high heat capacity H
- 4) Water has a high latent heat of vaporization, which is the amount of heat required to change water into steam, and it also has a high latent heat of fusion, which is the amount of heat needed to melt ice L

2. What is latent heat? What is sensible heat?

Latent heat- Heat added (or released) to change the phase (state) of water without a change in temp.

Sensible heat- Energy absorbed or released by substance that will change the temp of that substance

3. What is a calorie?

Amount of heat needed to raise the temperature of 1g of water by 1 deg. C

4. Using the graph on pg. 113 describe how many calories it takes to melt 1 gram of water. How many calories does it take to vaporize 1 gram of water? How many calories does it take to raise the temperature of 1 gram of water 100 degrees C?

To vaporize must have 540cals/g

To melt must have 80cals/g

To raise the temp of 1g of water 100 degrees C you need 100 cal.

Answer the above questions but for 5 grams of water.

$$540 \times 5 = 2,700 \text{ cal/g}$$

5. What is the hydrologic cycle? How is heat removed from the ocean? How is heat released to the atmosphere?

Hydrologic cycle- The general circulation of water at the earth's surface involving the processes of evaporation, condensation, precipitation, infiltration, and runoff.

Hydrologic cycle= the ultimate recycling system on Earth, controlling both the distribution of water and heat around the planet

Heat is removed from the ocean by evaporation as well as heat being released to the atmosphere.

6. What are the six most common ions in water? Which of these elements are derived from chemical weathering of rock? Which are from volcanic emissions? Which are from hydrothermal activity? Which are from fluid flow of accretionary prisms?

6 common ions include: Calcium 1.2%, Magnesium 3.7%, Potassium 1.1%, Sodium 30.6%, Sulfate 7.7%, and Chloride 55%. Volcanic emissions- CO₂, H₂O, HCL

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Table of Contents Oceanography Test Study Guide Answers

1. Understanding the eBook Oceanography Test Study Guide Answers
 - The Rise of Digital Reading Oceanography Test Study Guide Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Oceanography Test Study Guide Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Oceanography Test Study Guide Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Oceanography Test Study Guide Answers
 - Personalized Recommendations
 - Oceanography Test Study Guide Answers User Reviews and Ratings

- Oceanography Test Study Guide Answers and Bestseller Lists
- 5. Accessing Oceanography Test Study Guide Answers Free and Paid eBooks
 - Oceanography Test Study Guide Answers Public Domain eBooks
 - Oceanography Test Study Guide Answers eBook Subscription Services
 - Oceanography Test Study Guide Answers Budget-Friendly Options
- 6. Navigating Oceanography Test Study Guide Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Oceanography Test Study Guide Answers Compatibility with Devices
 - Oceanography Test Study Guide Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Oceanography Test Study Guide Answers
 - Highlighting and Note-Taking Oceanography Test Study Guide Answers
 - Interactive Elements Oceanography Test Study Guide Answers
- 8. Staying Engaged with Oceanography Test Study Guide Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Oceanography Test Study Guide Answers
- 9. Balancing eBooks and Physical Books Oceanography Test Study Guide Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Oceanography Test Study Guide Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Oceanography Test Study Guide Answers
 - Setting Reading Goals Oceanography Test Study Guide Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Oceanography Test Study Guide Answers
 - Fact-Checking eBook Content of Oceanography Test Study Guide Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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