

8. Buffer, passing through the, is shown within the Earth's upper. Buffer is composed and the composition of these all according to the.

12 samples of Buffer from a sampling were analysed to give the following composition of samples.

sample	composition (%)
1%	0.001
1%	0.01
1%	0.02

- (a) Define the term relative atomic mass.

The relative atomic mass of an element is the average mass of the atoms of the element, taking into account the relative abundance of the isotopes.

- (b) Calculate the relative atomic mass of the sample of buffer.

Give your answer to three decimal places.

$$\left(\frac{1}{100} \times 1.001 \right) + \left(\frac{1}{100} \times 1.01 \right) + \left(\frac{1}{100} \times 1.02 \right)$$

1.011

Answer = 1.011 (3 d.p.)

- (c) Some studies, on early, from, suggest, indicated that specimens were made up of any particular subset among which could be the distinct. However, evidence from of the presence of sub-atomic particles in, and in, is.

Complete the table to show the number of sub-atomic particles in the ^{12}C atom and $^{12}\text{C}^{2+}$ ion.

	protons	neutrons	electrons
^{12}C	6	6	6
$^{12}\text{C}^{2+}$	6	6	4



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Mike Smith

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'Mike Smith' text.

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