

# Griddable Science Items

## Topography map

## Density

## Atoms

$$F = ma$$

			6	.	2	5
0	0	0	0		0	0
1	1	1	1		1	1
2	2	2	2		2	2
3	3	3	3		3	3
4	4	4	4		4	4
5	5	5	5		5	5
6	6	6	6		6	6
7	7	7	7		7	7
8	8	8	8		8	8
9	9	9	9		9	9

Designed for .....

# 8th grade STAAR Test



but would work for other standardized tests


### Cricketables

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1. A balance and a graduated cylinder are used to determine the density of a mineral sample. The sample has a mass of 16.7 g and a volume of 14.2 mL. What is the density of the mineral sample in g/mL?


$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{16.7}{14.2} = 1.18 \text{ g/mL}$$

2. A rock is 1.2 cm<sup>3</sup>.  
What is the density of the rock in g/cm<sup>3</sup>?



$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{21.6}{1.2} = 18 \text{ g/cm}^3$$


3. A mineral grain has a mass of 10.0 g and a volume of 2.0 cm<sup>3</sup>. What is the density of the mineral grain in g/cm<sup>3</sup>?



$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{10.0}{2.0} = 5.0 \text{ g/cm}^3$$


Name: \_\_\_\_\_ ANSWERS

1. What is the density of the liquid in the graduated cylinder?




$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{4.00}{10.00} = 0.40 \text{ g/mL}$$


2. The sample of metal has a mass of 25 g. What is the density in g/cm<sup>3</sup>?




$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{25}{20} = 1.25 \text{ g/cm}^3$$

Answer each question using your CRICKETABLES. Be sure to use the correct units.

1.  10.0 g / 2.0 cm<sup>3</sup> = 5.0 g/cm<sup>3</sup>

2.  16.7 g / 14.2 mL = 1.18 g/mL

3.  21.6 g / 1.2 cm<sup>3</sup> = 18 g/cm<sup>3</sup>

# Design 81 ST

# Staar Science Problems Griddable

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A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the 'Show What You Know Publishing' text.

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