Molecular Geometry Chart

# of Electron Groups	Number of Lone Pairs	Electron Pair Arrangement	Molecular Geometry	Approximate Bond Angles
2	0	linear	180"	180°
3	o	trigonal planar	1300	120°
	1	bent		<120°
4	o	tetrahedral	3	109.5°
	1	trigonal pyramid	3000	<109.5° (~107°)
	2	bent	100.5	<109.5°(~105°)
5	o	trigonal bipyramidal	13-38-5	90°, 120°
	1	see-saw	12-3	<90°, <120°
	2	T-structure		<90°
	3	linear	() I 1 1 1 1 1 1 1 1 1	180°
6	o	octahehral	833.00	90°, 90°
	1	square pyramidal	990"	90°, <90°
	2	square planar	8:3:9:	90°

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Dale Atkins

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