Precipitate Lab

Procedure

- Locate a transparency sheet containing a data table. Be certain that the sheet is clean and dry.
- Place a drop of each of the two ionic compounds in each of the indicated boxes.
 Be careful not to touch the tip of the dropper bottle to the solutions. This will prevent contamination.
- For each combination note whether there is a precipitate and its color (e.g. WHITE PPT).
 If no reaction occurs (the mixed drop remain clear) write N.R. for no reaction or draw a dash —.
- If the precipitate is white, you may not be able to see it against the white background.
 Look at the drop sideways and/or slip a piece of colored paper below the droplets.

Data	® K₂CrO₄	⊘ CuSO₄	© Ba(OH)2	① Pb(NO ₃);	® Na ₂ CO ₃	Ф КСІ	© (NH ₄):003	
⊕ AgNO ₃	red-brown PPT	Foyn-voite	Ildini Assess	NP-	west ppr	TYP SPRUD	NA	
© (NH ₄)2€O ₃	NP-	High green	must bbs	ценай ррт	40	NA		
⊕ KCI	NP-	NR	LOVING PAT	mine the	NR		_	
⊕ Na ₂ CO ₃	NR	Figur value	N#	write PPT	Clean Up Rinse off the transparency with tap water, then with distilled water. (Tap water contains ions			
① Pb(NO ₃);	deron the	Sand Mar	WHILE					
© Ba(OH);	Yesau Pet	FRT WILL			that may change the data.) Dry the transparency with a paper towel.			
⊕ CuSO4	orange PPT							

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V. K. Sally, Chhaya Srivastava, Goyal Brothers Prakashan

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