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#### A-LEVEL CHEMISTRY

#### MARK SCHEME

Practice Paper 2

Maximum marks: 101

View detailed guidance on the conclusions you can draw from your students' performance in these papers on the MERiT welcome page. Understand how your students compare with others and target revision effectively by entering marks into MERiT.

Q1. (a) (i) M1 (Compounds / molecules with) the same structural formula

Penalise M1 if 'same structure' or 'different structural / displayed formula'.

M2 with atoms / bonds / groups arranged differently in space

OR atoms / bonds / groups with different spatial arrangements / different orientation

Ignore references to 'same molecular formula' or 'same empirical formula'.

Mark independently.

(0.0)

Credit C<sub>2</sub>H<sub>3</sub> Penalise C–CH<sub>3</sub>CH<sub>2</sub>

(b) Mit Br₂ OR bromine (water) OR bromine (in CCI₄ / organic solvent).

If M1, has no reagent or an incorrect reagent, CE=0. lance 'acidified'.

M2 Isomer 1: decolourised / goes colourless / loses its colour For M1 penalise Br (or incorrect formula of other correct reagent), but mark on.

M3 isomer 2: remains orange / red / yellow / brown / the same OR no reaction / 1 no (observable) change OR reference to colour going to the cyclopentane layer

For M1, it must be a whole reagent and / or correct formula.

If oxidation state given in name, it must be correct. If 'manganate' OR 'manganate(IV)' or incorrect formula, penalise M1, but mark on.

Alternatives : potassium manganate(VII)

M1 KMnO₁ in acid M2 colourless M3 purple

M1 KMnO<sub>4</sub> in alkali / neutral M2 brown solid M3 purple

Credit for the use of iodine

M1 iodine (solution / in KI) M2 colourless M3 (brown) to purple (credit no change)

Credit for the use of concentrated H<sub>2</sub>SO<sub>4</sub>

# **Predicted Paper 2 2014 November Mark Scheme**

**David Baud** 

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