

## 7-3 Review and Reinforcement

### Naming Chemical Compounds

Write the correct numerical prefix for each of the numbers in the table below.

Number	Prefix
1	mono
2	di
3	tri
4	tetra
5	penta
6	hexa
7	hepta
8	octa
9	nona
10	deca

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

cation  
anion

hydrates  
anhydrates

electronegative  
hydrogen ions

ionic  
molecular

- Ionic compounds that absorb water into their solid structures form substances called **—hydrate**.
- When naming an ionic substance, the name of the **cation** comes first.
- To name a binary molecular compound, add the suffix *-ide* to the more **electronegative** of the two elements.
- By one definition, an acid is a molecular compound that dissolves in water to produce **hydrogen ions**.
- The name of an acid is usually derived from the name of its characteristic **anion**.
- Prefixes are often used when naming **molecular** compounds.

# Prentice Hall Chemistry Answer Key 7

**L Towner**



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Many of the authors participated in a special symposium in Melbourne Australia honoring Kash Mittal's 100th edited book at the 18th Surfactants in Solution SIS meeting Each chapter provides an overview of a specific research area with discussions on past present and future directions The book is divided into six parts Part I reviews the evolution of theoretical models for surfactant self assembly and introduces a model for interpreting ion specific effects on aggregate properties Part II focuses on interactions of surfactant solutions with solid supports uses contact angles to understand hydrophobic hydrophilic changes in a lipid layer uses surface tension to understand molecular arrangements at interfaces reviews spreading phenomena discusses pattern formation on solid surfaces and applies tensiometry to probe flavor components of espresso Part III discusses novel DNA based materials multifunctional poly amino acid s based graft polymers for drug delivery and polymeric surfactants for stabilizing suspensions and emulsions Part IV introduces farm based biosurfactants from natural products and greener biosurfactants from bacteria Part V explores lyotropic liquid crystals and their applications in triggered drug release microemulsion properties and controlled drug release the role of hydrotopes in formulations and in enhancing solubilization in liquid crystals the potential of ionic liquids to generate tunable and selective reaction media and provides an overview of stimuli responsive surfactants Focusing on emulsions Part VI reviews the design of emulsion properties for various commercial applications the role of surfactants in the oil and gas industries and surfactant mechanisms for soil removal via microemulsions and emulsification

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The British National Bibliography Arthur James Wells, 1996

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