

## 9-2 Practice Problems (Continued)

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11. Write a balanced chemical equation for the reaction in which Dinitrogen Pentoxide reacts with water to produce Nitric Acid.
12. Magnesium reacts with Titanium (IV) Chloride to produce Magnesium Chloride and Titanium. Write the balanced equation for this reaction.
13. Write a balanced chemical equation for the reaction in which Carbon reacts with Zinc Oxide to produce Zinc and Carbon Dioxide.
14. Bromine reacts with Sodium Iodide to form Sodium Bromide and Iodine. Write the balanced equation for this reaction.
15. Write a balanced chemical equation for the reaction in which phosphorus Trichloride reacts with Chlorine gas to produce Phosphorus Pentachloride.
16. Phosphorus reacts with Bromine to produce Phosphorus Tribromide. Write the balanced equation for this reaction.
17. Calcium Hydride reacts with water to produce Calcium Hydroxide and Hydrogen gas. Write the balanced equation for this reaction.
18. Write a balanced chemical equation for the reaction in which Sulfuric Acid reacts with Potassium Hydroxide to produce Potassium Sulfate and water.
19. Write a balanced chemical equation for the reaction in which Propane ( $C_3H_8$ ) reacts with Oxygen gas to produce Carbon Dioxide and water.
20. Benzene ( $C_6H_6$ ) reacts with Oxygen gas to produce Carbon Dioxide and water. Write the balanced equation for this reaction.

# Prentice Hall Chemistry Nine Practice Problems Answer

**SB Merriam**

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

## **Prentice Hall Chemistry Nine Practice Problems Answer:**

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numerous updates have been made including Changes in naming and notation the first edition used the traditional names for the Gibbs Free Energy and for Partial Molal Properties this edition uses the more popular Gibbs Energy and Partial Molar Properties changes in symbols the first edition used the Lewis Randal fugacity rule and the popular symbol for the same quantity this edition only uses the popular notation and new problems have been added to the text Finally the second edition includes an appendix about the Bridgman table and its use

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