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## Ionic bonding practice

How many valence electrons do each of the following elements have?

- 1) carbon \_\_\_\_\_
- 2) selenium \_\_\_\_\_
- 3) xenon \_\_\_\_\_
- 4) potassium \_\_\_\_\_

Which of the following ions are likely to be formed? Circle one:

- |                    |     |    |                      |     |    |
|--------------------|-----|----|----------------------|-----|----|
| 5) $\text{N}^{+5}$ | yes | no | 8) $\text{Al}^{+2}$  | yes | no |
| 6) $\text{He}^{+}$ | yes | no | 9) $\text{P}^{-3}$   | yes | no |
| 7) $\text{F}^{-1}$ | yes | no | 10) $\text{Mg}^{+2}$ | yes | no |

Write the formula which result from combining the following ions:

- 11)  $\text{K}^{+1}$  and  $\text{Br}^{-}$  \_\_\_\_\_
- 12)  $\text{Li}^{+1}$  and  $\text{O}^{-2}$  \_\_\_\_\_
- 13)  $\text{Ca}^{+2}$  and  $\text{Cl}^{-1}$  \_\_\_\_\_
- 14)  $\text{Al}^{+3}$  and  $\text{S}^{-2}$  \_\_\_\_\_
- 15)  $\text{Fe}^{+2}$  and  $\text{S}^{-2}$  \_\_\_\_\_
- 16)  $\text{Na}^{+1}$  and  $\text{NO}_3^{-1}$  \_\_\_\_\_
- 17)  $\text{Na}^{+1}$  and  $\text{SO}_4^{-2}$  \_\_\_\_\_
- 18)  $\text{NH}_4^{+1}$  and  $\text{CO}_3^{-2}$  \_\_\_\_\_
- 19)  $\text{Fe}^{+3}$  and  $\text{CrO}_4^{-2}$  \_\_\_\_\_
- 20)  $\text{Al}^{+3}$  and  $\text{PO}_4^{-3}$  \_\_\_\_\_

# Skill Practice 19 Ionic Bonding Answer Key

**David R. Klein**



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