

Solubility Rules for Common Ionic Compounds in Water

Some Soluble Compounds

Group 1A cations (Li^+ , Na^+ , K^+ , Rb^+ , Cs^+), NH_4^+

NO_3^- , ClO_4^- , CH_3COO^- , ClO_3^-

Cl^- , Br^- , I^-

SO_4^{2-}

Exceptions

No Exceptions

No Exceptions

Halides of Ag^+ , Hg_2^{2+} , Pb^{2+}

Sulfates of Sr^{2+} , Ba^{2+} , Hg_2^{2+} , Pb^{2+}

Some Insoluble Compounds

S^{2-}

Sulfides of Group 1A cations, NH_4^+ , Ca^{2+} , Sr^{2+} , Ba^{2+}

CO_3^{2-}

Carbonates of Group 1A cations and NH_4^+

OH^-

Hydroxides of Group 1A cations, NH_4^+ , Ca^{2+} , Sr^{2+} , Ba^{2+}

PO_4^{3-}

Phosphates of Group 1A cations and NH_4^+

Precipitate Rule Sheet

Benjamin Clinton Kadel



Precipitate Rule Sheet:

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Bindeman, Andreas Pack, 2021-02-22 Volume 86 of Reviews in Mineralogy and Geochemistry concentrates on understanding the variations among ratios of the three isotopes of oxygen with primary emphasis on terrestrial systems Triple oxygen isotope variations may be related to large mass independent fractionation effects such observed in the Earth atmosphere or may be small and related to minute variations due to purely mass dependent processes Recent advancements in analytical resolution now allow for the identification of processes and distinct reservoirs that were formerly hidden in the paradigm of a single terrestrial fractionation line New high resolution measurements are accompanied by advances in theoretical calculations that dovetail with empirical calibrations and applications throughout this volume 14 chapters span a wide range of subjects from ab initio theoretical approaches to observation of triple oxygen isotope variations in the Earth litho hydro and atmosphere **Air Weather Service Technical Report** United States. Air Weather Service, 1970

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