

7-1 Review and Reinforcement

Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative
positive
crisscross

noble gas
electrons
anion

octet
ions
cation

electron configuration
polyatomic
empirical formula

1. An ionic bond is an attraction between oppositely charged ions.
2. Anions have a negative charge.
3. An atom becomes an ion by losing or gaining electrons.
4. The octet rule states that atoms tend to gain, lose, or share electrons in order to acquire a full set of valence electrons.
5. When sodium and chlorine form an ionic bond, both ions acquire the electron configuration of a(n) noble gas.
6. The atoms that make up polyatomic ions are bonded together by covalent bonds.
7. The crisscross method can be used to write the formula for an ionic compound.
8. The empirical formula of a compound denotes the ratio of ions in the compound.

If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line.

losing

true

true

polyatomic

seven

true (released)

9. Calcium becomes a monatomic cation by gaining two electrons.

10. A cation has a positive charge.

11. A binary ionic compound contains only one kind of cation and one kind of anion.

12. Monatomic ions consist of more than one atom.

13. The Lewis dot diagram for chlorine, a group 7A element, has six electrons.

14. A great deal of energy is produced when an ionic compound is formed.

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CO Houle

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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Unveiling the Power of Verbal Art: An Mental Sojourn through **Prentice Hall Chemistry 4 2 Review And Reinforcement**

In some sort of inundated with monitors and the cacophony of quick interaction, the profound energy and psychological resonance of verbal artistry frequently disappear in to obscurity, eclipsed by the continuous barrage of sound and distractions. However, set within the musical pages of **Prentice Hall Chemistry 4 2 Review And Reinforcement**, a interesting function of literary beauty that pulses with organic thoughts, lies an unforgettable trip waiting to be embarked upon. Written by way of a virtuoso wordsmith, this mesmerizing opus courses viewers on an emotional odyssey, lightly revealing the latent potential and profound impact embedded within the complicated web of language. Within the heart-wrenching expanse of the evocative examination, we will embark upon an introspective exploration of the book is key themes, dissect its interesting writing style, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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