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THE BEHAVIOR OF GASES

Chapter Test II

A. Matching

Match a word or phrase in Column I with the correct description in Column II. Write the letter of the correct answer next to the item.

Column I

1. The constant relating unit temperature changes to changes in the volume of a fixed amount of gas at constant pressure
2. The volume of a fixed amount of gas at constant pressure when the absolute temperature is 273 K
3. The ratio of the mass of a gas to its volume, independent of the actual amount of gas used
4. Temperature at which a gas would exert zero pressure if it cooled
5. A measure of how much the volume of a gas decreases when cooled
6. The combination of gas pressure, temperature, and volume of the gas mathematically expressed
7. An ideal gas that obeys the gas laws perfectly
8. $\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$
9. A measure of how much a gas expands per degree of heat
10. A gas at zero pressure

Column II

- a. standard pressure
- b. ideal gas constant R
- c. density
- d. extrapolation
- e. Boyle's law
- f. Boyle's equation
- g. Charles' law of expansion
- h. effusion
- i. compressibility
- j. Dalton's law of partial pressures

B. Multiple Choice

Choose the correct answer and write its letter on the line.

1. Diffusing the volume of a substance per unit time under steady temperature conditions is known as:
a. the diffusion coefficient
b. the rate of diffusion
c. the diffusion rate
d. the diffusion constant

Prentice Hall Chemistry Chapter 8 Assessment Answers

S Ben Porath



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