

13

ELECTRONS IN ATOMS

Conceptual Curriculum

- ☒ Concrete concepts
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Standard Curriculum

- ☒ Core content
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Honors Curriculum

- ☒ Core Honors content
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SECTION 13.1 MODELS OF THE ATOM (pages 361–366)

This section summarizes the development of atomic theory. It also explains the significance of quantized energies of electrons as they relate to the quantum mechanical model of the atom.

► The Evolution of Atomic Models (pages 361–363)

1. What are the chemical properties of atoms, ions, and molecules related to?

The properties are related to the arrangement of electrons.

2. Complete the table about atomic models and the scientists who developed them.

Scientist	Model of Atom
Dalton	The atom is a solid indivisible mass.
Thomson	The atom is a ball of positive charge with electrons stuck into the ball.
Rutherford	Most of an atom's mass is concentrated in the small, positively charged nucleus. The electrons surround the nucleus and the rest of the atom is empty space.
Bohr	Electrons are arranged in concentric circular paths around the nucleus.

3. The energy level of an electron is the region around the nucleus where the electron is likely to be moving.
4. Is the following sentence true or false? The electrons in an atom can exist between energy levels. false
5. Circle the letter of the term that completes the sentence correctly. A quantum of energy is the amount of energy required to
- move an electron from its present energy level to the next lower one
 - maintain an electron in its present energy level
 - ☒ move an electron from its present energy level to the next higher one
6. In general, the higher the electron is on the energy ladder, the farther it is from the nucleus.

Prentice Hall Chemistry 13 1 Worksheet Answers

A Gutmann



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KEY BENEFIT Physical Chemistry for the Life Sciences presents the core concepts of physical chemistry with mathematical rigor and conceptual clarity and develops the modern biological applications alongside the physical principles The traditional presentations of physical chemistry are augmented with material that makes these chemical ideas biologically relevant applying physical principles to the understanding of the complex problems of 21st century biology

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web the theme in this poem is how hypercritical us humans can be and that we might not be as perfect as we think we are silverstein uses easy and quite childish language in his poem he also uses hyperbole repetition personification rhyme alliteration and metaphor

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her parents she can't go to school i cannot go to school today said little peggy ann mckay i have the measles and the mumps
a gash a rash and purple bumps

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photo via instagram therapysmithk the pressure to adjust to other's opinions only increases as we get older perhaps you
haven't followed the career path your parents intended you don't blah blah blah

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web but time went by and the boy grew older and the tree was often alone then one day the boy came to the tree and the tree
said come boy come and climb up my trunk and swing from my branches and eat apples and play in my shade and be happy i
am too big to climb and play said

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web personification is a literary device found often in children's literature this is an effective use of figurative language
because personification relies on imagination for understanding of course readers know at a logical level that nonhuman
things cannot feel

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web his books have been commended for their appeal to both adults and children silverstein's poems are often darkly
humorous irreverent and populated with invented characters such as the bloath in where the sidewalk ends who dwells in
the undergrowth and feeds upon poets and tea

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something to identify yourself with no matter what your age is sheldon allan silverstein was born on september 25 1930 was
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