

The Three States of Matter

Grade 5 - Properties and Changes in Matter

Handout - Answers

1. Draw the particles of the three different states of matter (solid, liquid, and gas).

Solid	Liquid	Gas
		

2. Provide an example of each state of matter discussed:

Solid: **Any solid (e.g. Chair)** Liquid: **Any liquid (e.g. juice)** Gas: **Any gas (e.g. Oxygen)**

3. At what temperature does water turn into a solid? **0°C**
4. At what temperature does water turn into a gas? **100°C**
5. Based on the experiments conducted do you think all liquids turn to solid at the same temperature? Do you think all liquids turn to gas at the same temperature? Explain.
(Based on any experiment) **Different liquids turn to gas at different temperatures. An example would be that the sugar crystals turned back into a solid even though it was not 0°C or below.**
6. Once something is a solid, does it stay that way forever? Explain.
***Hint: Think about the ice cream experiment and use it as an example.**
No, it can change back to being a liquid (and possibly a gas). For example, we made ice cream (a solid) out of liquid cream but when I left it out too long some of it started to melt and turn back into a liquid
7. Challenge Question: If sugar is a solid then why does it take the shape of the container it is put in? What about with salt or sand?
Sugar (as well as salt and sand) is still a solid. Each crystal has a fixed shape and volume, does not flow under shear stress, does not take the shape of its container, and is incompressible. Many small solids (grains) moving together can sometimes act like a liquid (called granular flow).

Prentice Hall Chemistry States Of Matter Answers

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Prentice Hall Science Explorer: Teacher's ed, 2005 Matter , *Encyclopedia of Chemical Physics and Physical Chemistry* John H. Moore, Nicholas D. Spencer, 2023-07-03 The Encyclopedia of Physical Chemistry and Chemical Physics introduces possibly unfamiliar areas explains important experimental and computational techniques and describes modern endeavors The encyclopedia quickly provides the basics defines the scope of each subdiscipline and indicates where to go for a more complete and detailed explanation Particular attention has been paid to symbols and abbreviations to make this a user friendly encyclopedia Care has been taken to ensure that the reading level is suitable for the trained chemist or physicist The encyclopedia is divided in three major sections FUNDAMENTALS the mechanics of atoms and molecules and their interactions the macroscopic and statistical description of systems at equilibrium and the basic ways of treating reacting systems The contributions in this section assume a somewhat less sophisticated audience than the two subsequent sections At least a portion of each article inevitably covers material that might also be found in a modern undergraduate physical chemistry text METHODS the instrumentation and fundamental theory employed in the major spectroscopic techniques the experimental means for characterizing materials the instrumentation and basic theory employed in the study of chemical kinetics and the computational techniques used to predict the static and dynamic properties of materials APPLICATIONS specific topics of current interest and intensive research For the practicing physicist or chemist this encyclopedia is the place to start when confronted with a new problem or when the techniques of an unfamiliar area might be exploited For a graduate student in chemistry or physics the encyclopedia gives a synopsis of the basics and an overview of the range of activities in which physical principles are applied to chemical problems It will lead any of these groups to the salient points of a new field as rapidly as possible and gives pointers as to where to read about the topic in more detail

Prentice Hall Chemistry, 2000 The Liquid State and Its Electrical Properties E.E. Kunhardt, L.G. Christophorou, L.H. Luessen, 2012-12-06 As the various disciplines of science advance they proliferate and tend to become more esoteric Barriers of specialized terminologies form which cause scientists to lose contact with their colleagues and differences in points of view emerge which hinder the unification of knowledge among the various disciplines and even within a given discipline As a result the scientist and especially the student is in many instances offered fragmented glimpses of subjects that are fundamentally synthetic and that should be treated in their own right Such seems to be the case of the liquid state Unlike the other states of matter gases solids and plasmas the liquid state has not yet received unified treatment probably because it has been the least explored and remains the least understood state of matter Occasionally events occur which help remove some of the barriers that separate scientists and disciplines alike Such an event was the ASI on The Liquid State held this past July at the lovely Hotel Tivoli Sintra in the picturesque town of Sintra Portugal approximately 30 km northwest of Lisbon Since this broad a subject could not be covered in one Institute the focus of the ASI was on a theme that provided a common thread of

understanding for all in attendance the Electrical Properties of the Liquid State *Molecular Thermodynamics Of Electrolyte Solutions* Lloyd L Lee, 2008-07-08 The introductory textbook provides an update on electrolyte thermodynamics with a molecular perspective It is eminently suited as an introduction to the solution thermodynamics of ionic mixtures at the undergraduate and graduate level It is also invaluable for the understanding and design in the engineering of natural gas treating and adsorption refrigeration with electrolytes *Structure of Matter* Attilio Rigamonti, Pietro Carretta, 2007-04-29

Intended Audience Approach and Presentation This text is intended for a course of about fifteen weeks for undergraduate students It arises from the adaptation and the amendments to a text for a full year course in Structure of Matter written by one of the authors A R about thirty years ago At that time only a few if any textbooks having the suited form for introduction to basic quantum properties of atoms molecules and crystals in a comprehensive and interrelated way were available Along the last twenty years many excellent books pursuing the aforementioned aim have been published some of them are listed at the end of this preface Still there are reasons in our opinion to attempt a further text devoted to the quantum roots of condensed matter properties A practical aspect in this regard involves the organization of the studies in Physics after the huge scientific outburst of the various topics of fundamental and technological character in recent decades In most Universities there is now a first period of three or four years common to all the students and devoted to elementary aspects followed by a more advanced program in rather specialized fields of Physics *Phases of Matter and their Transitions* Gijsbertus de With, 2023-10-25

Phases of Matter and their Transitions An all in one comprehensive take on matter and its phase properties In Phases of Matter and their Transitions accomplished materials scientist Dr Gijsbertus de With delivers an accessible textbook for advanced students in the molecular sciences It offers a balanced and self contained treatment of the thermodynamic and structural aspects of phases and the transitions between them covering solids liquids gases and their interfaces The book lays the groundwork to describe particles and their interactions from the perspective of classical and quantum mechanics and compares phenomenological and statistical thermodynamics It also examines materials with special properties like glasses liquid crystals and ferroelectrics The author has included an extensive appendix with a guide to the mathematics and theoretical models employed in this resource Readers will also find Thorough introductions to classical and quantum mechanics intermolecular interactions and continuum mechanics Comprehensive explorations of thermodynamics gases liquids and solids Practical discussions of surfaces including their general aspects for solids and liquids Fulsome treatments of discontinuous and continuous transitions including discussions of irreversibility and the return to equilibrium Perfect for advanced students in chemistry and physics Phases of Matter and their Transitions will also earn a place in the libraries of students of materials science *Principles And Applications Of Positron And Positronium Chemistry* Jerry Y C Jean, Peter E Mallon, D M Schrader, 2003-04-04

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and researchers in the area of positron annihilation spectroscopy The contributing authors are authoritative scientists prominent in the frontiers of research actively pursuing positron annihilation research on chemical and applied systems

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Kenneth S. Pitzer,2018-05-04 This book was first published in 1991 It considers the concepts and theories relating to mostly aqueous systems of activity coefficients *Catalog of Copyright Entries. Third Series* Library of Congress. Copyright

Office,1976 **Practical Chemical Thermodynamics for Geoscientists** Bruce Fegley Jr.,2012-10-22 Practical Chemical Thermodynamics for Geoscientists covers classical chemical thermodynamics and focuses on applications to practical problems in the geosciences environmental sciences and planetary sciences This book will provide a strong theoretical foundation for students while also proving beneficial for earth and planetary scientists seeking a review of thermodynamic principles and their application to a specific problem Strong theoretical foundation and emphasis on applications Numerous worked examples in each chapter Brief historical summaries and biographies of key thermodynamicists including their fundamental research and discoveries Extensive references to relevant literature Scientific, Medical, and Technical

Books Published in the United States of America Reginald Robert Hawkins, 1950 The Science Teacher , 1992 Some issues are accompanied by a CD ROM on a selected topic **Chemical Engineering Design** Ray Sinnott, Gavin Towler, 2009-05-15 Chemical Engineering Design is one of the best known and most widely adopted texts available for students of chemical engineering It completely covers the standard chemical engineering final year design course and is widely used as a graduate text The hallmarks of this renowned book have always been its scope practical emphasis and closeness to the curriculum That it is written by practicing chemical engineers makes it particularly popular with students who appreciate its relevance and clarity Building on this position of strength the fifth edition covers the latest aspects of process design operations safety loss prevention and equipment selection and much more Comprehensive in coverage exhaustive in detail and supported by extensive problem sets at the end of each chapter this is a book that students will want to keep to hand as they enter their professional life The leading chemical engineering design text with over 25 years of established market leadership to back it up an essential resource for the compulsory design project all chemical engineering students take in their final year A complete and trusted teaching and learning package the book offers a broader scope better curriculum coverage more extensive ancillaries and a more student friendly approach at a better price than any of its competitors Endorsed by the Institution of Chemical Engineers guaranteeing wide exposure to the academic and professional market in chemical and process engineering *Geological Survey Bulletin* , 1961 Natural Radioactive Disequilibrium of the Uranium Series John Nicholas Rosholt, 1959 This report concerns work done on behalf of the U S Atomic Energy Commission and is published with the permission of the Commission

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