

7-1 Review and Reinforcement (continued)

Draw the Lewis dot diagram for each of the following elements.

15. B



16. Cl



17. O



18. Ne



19. Si



20. Na



21. Al



22. Mg



Write the chemical formula for each of the following ionic compounds.

\_\_\_  $\text{Al}_2\text{S}_3$

\_\_\_ 23. aluminum sulfide

\_\_\_  $(\text{NH}_4)_2\text{CO}_3$

\_\_\_ 24. ammonium carbonate

\_\_\_  $\text{CaO}$

\_\_\_ 25. calcium oxide

\_\_\_  $\text{SrCl}_2$

\_\_\_ 26. strontium chloride

\_\_\_  $\text{K}_2\text{O}$

\_\_\_ 27. potassium oxide

\_\_\_  $\text{NaF}$

\_\_\_ 28. sodium fluoride

# Prentice Hall Chemistry Unit Test Answers

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## Table of Contents Prentice Hall Chemistry Unit Test Answers

1. Understanding the eBook Prentice Hall Chemistry Unit Test Answers
  - The Rise of Digital Reading Prentice Hall Chemistry Unit Test Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Prentice Hall Chemistry Unit Test Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Prentice Hall Chemistry Unit Test Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Prentice Hall Chemistry Unit Test Answers
  - Personalized Recommendations
  - Prentice Hall Chemistry Unit Test Answers User Reviews and Ratings
  - Prentice Hall Chemistry Unit Test Answers and Bestseller Lists
5. Accessing Prentice Hall Chemistry Unit Test Answers Free and Paid eBooks
  - Prentice Hall Chemistry Unit Test Answers Public Domain eBooks
  - Prentice Hall Chemistry Unit Test Answers eBook Subscription Services
  - Prentice Hall Chemistry Unit Test Answers Budget-Friendly Options
6. Navigating Prentice Hall Chemistry Unit Test Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Prentice Hall Chemistry Unit Test Answers Compatibility with Devices
  - Prentice Hall Chemistry Unit Test Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Prentice Hall Chemistry Unit Test Answers
  - Highlighting and Note-Taking Prentice Hall Chemistry Unit Test Answers
  - Interactive Elements Prentice Hall Chemistry Unit Test Answers
8. Staying Engaged with Prentice Hall Chemistry Unit Test Answers



- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Prentice Hall Chemistry Unit Test Answers
- 9. Balancing eBooks and Physical Books Prentice Hall Chemistry Unit Test Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Prentice Hall Chemistry Unit Test Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Prentice Hall Chemistry Unit Test Answers
  - Setting Reading Goals Prentice Hall Chemistry Unit Test Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Prentice Hall Chemistry Unit Test Answers
  - Fact-Checking eBook Content of Prentice Hall Chemistry Unit Test Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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