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# The Audco Manual

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## **The Audco Manual:**

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 $\text{Cl}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{HCl} + \text{O}_2$   
 $\text{Pb} + 2\text{HCl} \rightarrow \text{PbCl}_2 + \text{H}_2$   
 $2\text{Al} + 3\text{Br}_2 \rightarrow 2\text{AlBr}_3$   
 $\text{K}_2\text{SO}_4 + 2\text{H}_2\text{O} \rightarrow 2\text{KOH} + \text{H}_2\text{SO}_4$   
 $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$   
 $\text{C}_3\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$

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 $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$   
 $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$   
 $\text{RbH} + \text{H}_2\text{O} \rightarrow \text{RbOH} + \text{H}_2$   
 $2\text{Cl}_2 + 2\text{KBr} \rightarrow 2\text{KCl} + \text{Br}_2$   
 $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$   
 $3\text{Ca} + 2\text{N}_2 \rightarrow \text{Ca}_3\text{N}_2$   
 $2\text{CH}_4 + 4\text{H}_2\text{O} \rightarrow 2\text{CO}_2 + 8\text{H}_2$   
 $2\text{H}_2\text{O} + 2\text{H}_2\text{O} \rightarrow 2\text{H}_2\text{O} + 2\text{H}_2\text{O}$

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 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF}$   
 $\text{Cl}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{HCl} + \text{O}_2$   
 $\text{Pb} + 2\text{HCl} \rightarrow \text{PbCl}_2 + \text{H}_2$   
 $2\text{Al} + 3\text{Br}_2 \rightarrow 2\text{AlBr}_3$   
 $\text{K}_2\text{SO}_4 + 2\text{H}_2\text{O} \rightarrow 2\text{KOH} + \text{H}_2\text{SO}_4$   
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 $2\text{FeBr}_3 + 3\text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + 3\text{HBr}$   
 $\text{C}_4\text{H}_{10} + 13\text{O}_2 \rightarrow 4\text{CO}_2 + 10\text{H}_2\text{O}$

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 $3\text{Fe} + 2\text{O}_2 \rightarrow \text{Fe}_3\text{O}_4$   
 $2\text{Fe} + 3\text{CO} \rightarrow \text{Fe}_2\text{O}_3 + 3\text{C}$   
 $4\text{BaO} + 3\text{PO}_4 \rightarrow 2\text{Ba}_3(\text{PO}_4)_2 + \text{P}_2\text{O}_5$   
 $2\text{NH}_3 + 3\text{H}_2\text{O} \rightarrow 2\text{NH}_4\text{OH}$   
 $2\text{Al} + 3\text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + 3\text{H}_2$   
 $3\text{H}_2\text{O} + 2\text{H}_2\text{O} \rightarrow 3\text{H}_2\text{O} + 2\text{H}_2\text{O}$

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