



Figure 1.1: (a) Charges  $q_1$  and  $q_2$  have the same sign; electric force is repulsive. (b) Charges  $q_1$  and  $q_2$  have opposite signs; electric force is attractive.

for historical reasons but also because in later applications the constant  $\epsilon_0$  is more convenient.  $\epsilon_0$  is called the **permittivity constant**.<sup>3</sup>

When several point charges are present, the total force on a particular charge  $q_0$  is the **vector sum** of the individual forces gotten from Coulomb's law. (Thus, electric forces have a **superposition** property.) For a continuous distribution of charge we need to divide up the charge distribution into infinitesimal pieces and add up the individual forces with integrals to get the net force.

## 1.2 Worked Examples

### 1.2.1 Electric Charge

#### 1. What is the total charge of 75.0 kg of electrons? [HWs 23-19]

The mass of *one* electron is  $9.11 \times 10^{-31}$  kg, so that a mass  $M = 75.0$  kg contains

$$N = \frac{M}{m_e} = \frac{(75.0 \text{ kg})}{(9.11 \times 10^{-31} \text{ kg})} = 8.23 \times 10^{31} \quad \text{electrons}$$

The charge of *one* electron is  $-e = -1.60 \times 10^{-19}$  C, so that the total charge of  $N$  electrons is:

$$Q = N(-e) = (8.23 \times 10^{31})(-1.60 \times 10^{-19} \text{ C}) = -1.32 \times 10^{13} \text{ C}$$

2. (a) How many electrons would have to be removed from a penny to leave it with a charge of  $+1.0 \times 10^{-7}$  C? (b) To what fraction of the electrons in the penny does this correspond? [A penny has a mass of 3.11 g; assume it is made entirely of copper.] [HWs 23-23]

<sup>3</sup>In these notes,  $k$  will be used mainly in the first chapter; thereafter, we will make increasing use of  $\epsilon_0$ !

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**G Orfield**



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