

11.1

DESCRIBING CHEMICAL REACTIONS

Section Review

Objectives

1. Explain how a chemical equation is written.
2. Explain how to write a balanced equation.
3. Write the steps for writing a balanced chemical equation.

Vocabulary

- | | |
|----------------------|----------------------|
| 1. chemical equation | 4. coefficient |
| 2. stoichiometric | 5. balanced equation |
| 3. reactant | |

Part A Completion

Use the vocabulary words in each sentence, writing either the correct word form or the correct chemical equation. Write each answer on a separate sheet of paper, showing your work.

1. A chemical equation that includes information about the state of _____ is called a _____.
2. The chemical formulas of chemical _____ are the _____ of the reactants. The chemical formulas of the products are the _____.
3. The law of conservation of _____ states that matter is neither created nor destroyed in a chemical reaction.
4. The _____ of a chemical equation is the number of _____ of each element on the reactant side of the equation, which must equal the number of _____ of each element on the product side of the equation.
5. The _____ of a chemical equation is the number of _____ of each element on the reactant side of the equation, which must equal the number of _____ of each element on the product side of the equation.

6. The _____ of a chemical equation is the number of _____ of each element on the reactant side of the equation, which must equal the number of _____ of each element on the product side of the equation.
7. The _____ of a chemical equation is the number of _____ of each element on the reactant side of the equation, which must equal the number of _____ of each element on the product side of the equation.
8. The _____ of a chemical equation is the number of _____ of each element on the reactant side of the equation, which must equal the number of _____ of each element on the product side of the equation.

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Anthea Maton



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