#### Review for Ch. 8 Test

#### 8.1 Describing Chemical Reactions

 know how to write <u>word equations</u> (i.e. you have be able to name compounds, including acids, ionic, and covalent compounds from Ch. 7)

- know how to write a formula equation

 What 2 things do coefficients represent in an equation? Relative number of moles of a substance; relative number of molecules needed for a reaction to occur

- know common symbols used in a reaction

- know 4 evidences of a chemical reaction:

1. energy in the form of heat of light

2. production of a gas

3. change in color

4. formation of a precipitate

 know the <u>Law of conservation of matter</u>; matter can neither be created nor destroyed in a chemical reaction (which means, mass of reactants must equal the mass of products)

know what a chemical equation represents: relative amounts of reactants and products

know how to balance chemical equations (combustion reactions are particularly tricky, practice!)

Complete the equation and write a word equation for the following reaction: ZnO + FeSO₄ ⇒

ZnO + FeSO<sub>4</sub>  $\Longrightarrow$  ZnSO<sub>4</sub> + FeO

Zinc oxide reacts with iron (II) sulfate to yield zinc sulfate and iron (II) oxide

2. Write a formula equation for the reaction of a solution of AgCl with potassium metal:

AgCl (aq) + K (s) - KCl (aq) + Ag (s)

3. What elements are always diatomic gases at room temperature?

All of the halogens (Group 17), Hz, Nz, O2

4. Can you balance these reactions:

a) 2 C<sub>2</sub>H<sub>6</sub>(g) + 7 O<sub>2</sub>(g)  $\Longrightarrow$  4 CO<sub>2</sub>(g) + 6 H<sub>2</sub>O(g)

b) 2 Cu<sub>2</sub>S (s) + 3 O<sub>2</sub> (g) = 2 Cu<sub>2</sub>O (s) + 2 SO<sub>2</sub> (g)

c) 2 Al(s) + 3 Pb(NO<sub>3</sub>)<sub>2</sub> (aq)  $\Longrightarrow$  2 Al(NO<sub>3</sub>)<sub>3</sub> (aq) + 3 Pb(s)

# **Section 1describing Chemical Reaction Answer Key**

**RD Boyd** 

**Section 1describing Chemical Reaction Answer Key:** 

This book delves into Section 1describing Chemical Reaction Answer Key. Section 1describing Chemical Reaction Answer Key is a vital topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Section 1describing Chemical Reaction Answer Key, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
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  - Chapter 2: Essential Elements of Section 1describing Chemical Reaction Answer Key
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  - Chapter 4: Section 1describing Chemical Reaction Answer Key in Specific Contexts
  - ∘ Chapter 5: Conclusion

Key.

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- 6. In chapter 5, the author will draw a conclusion about Section 1describing Chemical Reaction Answer Key. The final chapter will summarize the key points that have been discussed throughout the book.

  This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Section 1describing Chemical Reaction Answer

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