

Quiz & Worksheet - Properties of Gases with the Kinetic Molecular Theory

1. Which of the following is TRUE about most gases?

- ☐ They expand to completely fill their container.
- ☐ They cannot be compressed.
- ☐ All of their particles are the same size.
- ☐ They have a fixed volume.
- ☐ Their particles are very close together.

2. According to the kinetic molecular theory, temperature is directly proportional to ____.

- ☐ average kinetic energy
- ☐ lattice energy
- ☐ ionization energy
- ☐ free energy
- ☐ average potential energy

3. What does it mean when a collision is elastic?

- ☐ No energy is gained or lost.
- ☐ Energy is gained.
- ☐ Energy is lost.
- ☐ The particles can stretch out.
- ☐ The particles slow down.

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