

Guided Practice - Acids & Bases**Properties of Acids and Bases**

1. Match the following terms with their descriptions:

- | | | |
|------------------------|-------|--|
| a. Arrhenius Acid | _____ | the substance formed when a proton is added to _____ |
| b. Arrhenius Base | _____ | _____ a base |
| c. Conjugate Acid | _____ | the remaining substance when a proton is lost from an acid |
| d. Conjugate Base | _____ | a substance that produces OH^- in solution |
| e. Bronsted-Lowry Acid | _____ | a substance producing H^+ in solution |
| f. Bronsted-Lowry Base | _____ | a proton acceptor |
| | | a proton donor |

2. Which of the following represent conjugate acid-base pairs? Label each species as a conjugate acid or a conjugate base. For those pairs that are H_2O conjugates, write the correct conjugate acid or base for each species in the pair, labeling each species correctly.

- HSO_4^- , SO_4^{2-}
- HBr , Br^-
- H_3PO_4 , PO_4^{3-}
- H_2O , HO^-

3. Complete the following table regarding acid strength.

Property	Strong Acid	Weak Acid
Definition		
Describe Dissociation		
Species present in solution		
Relative strength of conjugate base compared to water		
Equation showing acid's reaction to water		

Section 19 Acids And Bases Answer Key

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Section 19 Acids And Bases Answer Key:

Chemistry Therald Moeller, 2012-12-02 Chemistry with Inorganic Qualitative Analysis is a textbook that describes the application of the principles of equilibrium represented in qualitative analysis and the properties of ions arising from the reactions of the analysis. This book reviews the chemistry of inorganic substances as the science of matter, the units of measure used, atoms, atomic structure, thermochemistry, nuclear chemistry, molecules and ions in action. This text also describes the chemical bonds, the representative elements, the changes of state, water and the hydrosphere, which also covers water pollution and water purification. Water purification occurs in nature through the usual water cycle and by the action of microorganisms. The air flushes dissolved gases and volatile pollutants when water seeps through the soil; it filters solids as they settle in the bottom of placid lakes. Microorganisms break down large organic molecules containing mostly carbon, hydrogen, nitrogen, oxygen, sulfur, or phosphorus into harmless molecules and ions. This text notes that natural purification occurs if the level of contaminants is not so excessive. This textbook is suitable for both chemistry teachers and students.

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First Year Chemistry Students' Conceptions of Acid/base Chemistry Sally Diane Rupert,2001

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