

Study Guide Section 11.4 Percent Yield**Key Concepts**

- The theoretical yield of a chemical reaction is the maximum amount of product that can be produced from a given amount of reactant. Theoretical yield is calculated from the balanced chemical equation.
- The actual yield is the amount of product produced. Actual yield must be obtained through experimentation.
- Percent yield is the ratio of actual yield to theoretical yield expressed as a percent. High percent yield is important in reducing the cost of every product produced through chemical processes.

$$\text{percent yield} = \frac{\text{actual yield}}{\text{theoretical yield}} \times 100$$



Section 12 4 Percent Yield Study Guide For Content Mastery

Sue Ghazala



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The CRC Master Keyword Guide for Food Sue Ghazala, 2003-11-25 Although easily available and searchable on line the CFR 21 is a vast document covering a wide range of subjects but contains no index And sifting through the results of a simple search does not always provide the information you need in the context you need it After years of frustration you may have tried to construct your own index only to have it fail

Resources in Education ,2000-10 **Report** Pacific Northwest Forest and Range Experiment Station (Portland, Or.),1962

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