

## Oxford Resources for IB

## Chemistry – 2023 Edition

## Answers

## Reactivity 1.1 – Measuring enthalpy changes

## Activity (page 390)

- The glass of water is an open system.
- The system includes the water and ice in the glass. The surroundings are all matter that exists outside the glass of iced water. The glass can be considered the boundary between the system and the surroundings.
- Heat,  $Q$ , is a form of energy that is transferred from a warmer body to a cooler body, as a result of the temperature gradient. The water in the glass is warmer than the ice cubes ( $\text{H}_2\text{O}(\text{s})$ ). Energy will flow from the water into the ice cubes, warming them, so eventually they will melt. As the surroundings are warmer than the iced water, energy will flow from the surroundings into the system.
- Condensation occurs on the outside of the glass and you observe liquid water. The change of state is from a gas (water vapour) to a liquid (water). Energy is transferred into the surroundings when water vapour condenses into liquid water.

## Practice question (page 393)

1. C

## Practice questions (page 394)

2. a.  $Q = mc\Delta T$   
 $= 1 \text{ kg} \times 4.18 \text{ kJ kg}^{-1} \text{ K}^{-1} \times 1 \text{ K}$   
 $= 4.18 \text{ kJ}$
- b.  $Q = mc\Delta T$   
 $= 1000 \text{ kg} \times 0.385 \text{ kJ kg}^{-1} \text{ K}^{-1} \times 1 \text{ K}$   
 $= 385 \text{ kJ}$

3. a

4. a

## Practice questions (page 396)

5.  $Q = mc\Delta T$   
 $= 30 \text{ g} \times 4.18 \text{ J g}^{-1} \text{ K}^{-1} \times 30 \text{ K}$   
 $= 3762 \text{ J}$
6. First, convert kJ to J ( $\times 1000$ )  
Next, convert g to kg ( $\div 1000$ )  
 $Q = mc\Delta T$   
 $\Delta T = \frac{Q}{mc}$   
 $= \frac{675 \text{ J}}{0.125 \text{ kg} \times 385 \text{ J kg}^{-1} \text{ K}^{-1}}$   
 $= 14.0 \text{ K (3 sf)}$



# Plato Answers Chemistry Ib

**Sebastian Brünink**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## **Plato Answers Chemistry Ib:**

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## **Table of Contents Plato Answers Chemistry Ib**

1. Understanding the eBook Plato Answers Chemistry Ib
  - The Rise of Digital Reading Plato Answers Chemistry Ib
  - Advantages of eBooks Over Traditional Books
2. Identifying Plato Answers Chemistry Ib
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Plato Answers Chemistry Ib
  - User-Friendly Interface
4. Exploring eBook Recommendations from Plato Answers Chemistry Ib
  - Personalized Recommendations
  - Plato Answers Chemistry Ib User Reviews and Ratings
  - Plato Answers Chemistry Ib and Bestseller Lists
5. Accessing Plato Answers Chemistry Ib Free and Paid eBooks
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  - Plato Answers Chemistry Ib eBook Subscription Services
  - Plato Answers Chemistry Ib Budget-Friendly Options
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  - Plato Answers Chemistry 1b Enhanced eBook Features
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    - Highlighting and Note-Taking Plato Answers Chemistry 1b
    - Interactive Elements Plato Answers Chemistry 1b
  8. Staying Engaged with Plato Answers Chemistry 1b
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Plato Answers Chemistry 1b
  9. Balancing eBooks and Physical Books Plato Answers Chemistry 1b
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Plato Answers Chemistry 1b
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Plato Answers Chemistry 1b
    - Setting Reading Goals Plato Answers Chemistry 1b
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Plato Answers Chemistry 1b
    - Fact-Checking eBook Content of Plato Answers Chemistry 1b
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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