

Name _____

Chemistry Worksheet
Naming & Formula Writing (Ionic)

Instructions: Write the formulas &/or the names for the compounds listed below

- | | | | |
|--------------------------|-------|--|-------|
| 1. Sodium nitrate | _____ | 26. Aluminum chloride | _____ |
| 2. Calcium carbonate | _____ | 27. Iron (III) hydroxide | _____ |
| 3. Magnesium oxide | _____ | 28. Sodium acetate | _____ |
| 4. Ammonium sulfide | _____ | 29. calcium hydroxide | _____ |
| 5. Lead (II) sulfate | _____ | 30. sodium iodate | _____ |
| 6. Sodium cyanide | _____ | 31. Nickel (II) nitrate | _____ |
| 7. Potassium hydroxide | _____ | 32. Iron (II) chloride | _____ |
| 8. Silver chloride | _____ | 33. Magnesium bromide | _____ |
| 9. Iron (III) hydroxide | _____ | 34. Ammonium nitrate | _____ |
| 10. Potassium hydroxide | _____ | 35. Silver bromide | _____ |
| 11. Tin (IV) perchlorate | _____ | 36. $\text{Al}(\text{OH})_3$ | _____ |
| 12. Potassium carbonate | _____ | 37. NH_4I | _____ |
| 13. Silver nitrate | _____ | 38. Li_2CO_3 | _____ |
| 14. Sodium iodide | _____ | 39. CuSO_4 | _____ |
| 15. Ammonium hydroxide | _____ | 40. KCN | _____ |
| 16. Potassium iodate | _____ | 41. $\text{Pb}(\text{ClO})_2$ | _____ |
| 17. Lead (IV) oxide | _____ | 42. BaS | _____ |
| 18. Ammonium hydroxide | _____ | 43. ZnSO_4 | _____ |
| 19. Barium sulfate | _____ | 44. $\text{Pb}(\text{CH}_3\text{COO})_2$ | _____ |
| 20. barium chloride | _____ | 45. $\text{Ca}(\text{NO}_3)_2$ | _____ |
| 21. Cobalt (II) chloride | _____ | 46. $\text{Fe}_2(\text{CO}_3)_3$ | _____ |
| 22. Sodium carbonate | _____ | 47. NH_4IO_3 | _____ |
| 23. Calcium oxide | _____ | 48. CaCl_2 | _____ |
| 24. Lead (II) nitrate | _____ | 49. NaF | _____ |
| 25. Tin (II) chloride | _____ | 50. $\text{Cu}(\text{NO}_3)_2$ | _____ |

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Peter Atkins, Sapling Learning



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Table of Contents Sapling Learning Answers Chemistry Names And Formulas

1. Understanding the eBook Sapling Learning Answers Chemistry Names And Formulas
 - The Rise of Digital Reading Sapling Learning Answers Chemistry Names And Formulas
 - Advantages of eBooks Over Traditional Books
2. Identifying Sapling Learning Answers Chemistry Names And Formulas
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sapling Learning Answers Chemistry Names And Formulas
 - User-Friendly Interface
4. Exploring eBook Recommendations from Sapling Learning Answers Chemistry Names And Formulas

- Personalized Recommendations
- Sapling Learning Answers Chemistry Names And Formulas User Reviews and Ratings
- Sapling Learning Answers Chemistry Names And Formulas and Bestseller Lists
- 5. Accessing Sapling Learning Answers Chemistry Names And Formulas Free and Paid eBooks
 - Sapling Learning Answers Chemistry Names And Formulas Public Domain eBooks
 - Sapling Learning Answers Chemistry Names And Formulas eBook Subscription Services
 - Sapling Learning Answers Chemistry Names And Formulas Budget-Friendly Options
- 6. Navigating Sapling Learning Answers Chemistry Names And Formulas eBook Formats
 - ePub, PDF, MOBI, and More
 - Sapling Learning Answers Chemistry Names And Formulas Compatibility with Devices
 - Sapling Learning Answers Chemistry Names And Formulas Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sapling Learning Answers Chemistry Names And Formulas
 - Highlighting and Note-Taking Sapling Learning Answers Chemistry Names And Formulas
 - Interactive Elements Sapling Learning Answers Chemistry Names And Formulas
- 8. Staying Engaged with Sapling Learning Answers Chemistry Names And Formulas
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sapling Learning Answers Chemistry Names And Formulas
- 9. Balancing eBooks and Physical Books Sapling Learning Answers Chemistry Names And Formulas
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sapling Learning Answers Chemistry Names And Formulas
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Sapling Learning Answers Chemistry Names And Formulas
 - Setting Reading Goals Sapling Learning Answers Chemistry Names And Formulas
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Sapling Learning Answers Chemistry Names And Formulas

- Fact-Checking eBook Content of Sapling Learning Answers Chemistry Names And Formulas
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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