Chemical Equations and Reactions

SECTION 2

SHORT ANSWER Answer the following questions in the space provided.

1. Match the equation type on the left to its representation on the right,

___ synthesis

(a) AX + BY → AY + BX

___d ___ decomposition

(b) $A + BX \rightarrow AX + B$

b single-displacement

(c) A + B → AX (d) AX → A + X

- _a__ double-displacement
- In the reaction described by the equation 2Al(s) + 3Fe(NO₅)₂(aq) → 3Fe(s) + 2Al(NO₅)₂(aq), iron has been replaced by
 - (a) nitrate. (c) aluminum.
 - (b) water.

(d) nitrogen.

- Of the following chemical equations, the only reaction that is both synthesis and combustion is
 - (a) C(s) + O₂(g) → CO₂(g).
 - (b) $2C_4H_{10}(I) + 13O_2(g) \rightarrow 8CO_2(g) + 10H_2O(I)$.
 - (c) $6CO_2(g) + 6H_2O(g) \rightarrow C_6H_{12}O_6(aq) + 6O_2(g)$.
 - (d) $C_6H_{12}O_6(aq) + 6O_2(g) \rightarrow 6CO_2(aq) + 6H_2O(l)$.
- __b__ Of the following chemical equations, the only reaction that is both combustion and decomposition is
 - (a) S(s) + O₂(g) → SO₂(g).
 - (b) 2C₄H₁₀(I) + 13O₂(g) → 8CO₂(g) + 10H₂O(I).
 - (c) $2H_2O_2(I) \rightarrow 2H_2O(I) + O_2(g)$.
 - (d)
- Identify the products when the following substances decompose:
 - a. Its separate elements _____ a binary compound
 - b. Metal oxide + water _____ most metal hydroxides
 - c. Metal oxide + carbon dioxide a metal carbonate
 - d. Water + sulfur dioxide the acid H₂SO₃
- The complete combustion of a hydrocarbon in excess oxygen yields the products

Quantities In Chemical Reactions Section Review Answers

Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Dalius Sagatys, Adrian George

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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001

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