

# Chemistry Form 5 Chapter 1

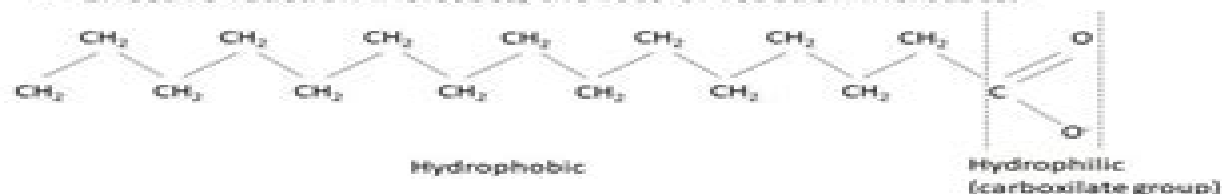
## Rate of Reaction

- Rate of reaction is the change in quantity of reactant or product against time
- Method used to measure the volume of hydrogen gas is **water displacement method**

| <u>Fast reaction</u>              | <u>Slow reaction</u>             |
|-----------------------------------|----------------------------------|
| Short time taken for the reaction | Long time taken for the reaction |
| High rate of reaction             | Low rate of reaction             |
| Sodium + Water                    | Copper(II) oxide + Hydrogen gas  |
| Neutralisation                    | Fermentation                     |
| Precipitation                     | Rusting                          |

## Effective reaction

- Effective reaction are **collisions** between the **particles of reactants** that cause a reaction
- Occur when the particles of reactants have energy that is equal or higher than **activation energy**.
- Effective reaction increases, the rate of reaction increases.



## Graph

# Question General Chemistry Form5

**Alexander Stone Macnow**



### **Question General Chemistry Form5:**

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**EBOOK: GENERAL CHEMISTRY, THE ESSENTIAL CONCEPTS** CHANG, 2013-01-07 **EBOOK GENERAL CHEMISTRY THE ESSENTIAL CONCEPTS** **General Chemistry Workbook** Daniel C. Tofan, 2010-07-28 This workbook is a

comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions. It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course. It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material.

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**Problem Exercises for General Chemistry** George Gilbert Long, Forrest C. Hentz, 1982

Understanding Basic Chemistry Through Problem Solving: The Learner's Approach (Revised Edition) Kim Seng Chan, Jeanne Tan, 2017-03-02 This book is the revised edition of Understanding Basic Chemistry Through Problem Solving published in 2015 It is in a series of Understanding Chemistry books which deals with Basic Chemistry using the problem solving approach Written for students taking either the university of Cambridge O level examinations or the GCSE examinations this guidebook covers essential topics and concepts under both stipulated chemistry syllabi The book is written in such a way as to guide the reader through the understanding and applications of essential chemical concepts using the problem solving approach The authors have also retained the popular discourse feature from their previous few books Understanding Advanced Physical Inorganic Chemistry Understanding Advanced Organic and Analytical Chemistry Understanding Advanced Chemistry Through Problem Solving and Understanding Basic Chemistry to help the learners better understand and see for themselves how the concepts should be applied during solving problems Based on the Socratic Method questions are implanted throughout the book to help facilitate the reader's development in forming logical conclusions of concepts and the way they are being applied to explain the problems In addition the authors have also included important summaries and concept maps to help the learners to recall remember reinforce and apply the fundamental chemical concepts in a simple way

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**Problem-solving in General Chemistry** Christopher John Willis, 1977

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