



Qualitative Analysis and Chemical Bonding

Concepts:

Chemical bonds, structure-property relationships, ionic bonds, covalent bonds, polar covalent bonds, nonpolar covalent bonds, metallic bonds

Use the following recommendations to increase and/or decrease the challenge difficulty for your students.

Short-on-time Inquiry Lab:

Students follow a detailed procedure to test for six properties of six compounds to determine the type of bonding or interparticle force that holds the constituent particles in each compound together. Students answer questions meant to probe their understanding of the structure-property relationship. (30 min.)

Guided Inquiry Lab:

Students follow a detailed procedure to test for six properties of twelve compounds to determine the type of bonding or interparticle force that holds the constituent particles in each compound together. Students answer questions meant to probe their understanding of the structure-property relationship. (50 min.)

Open Inquiry Lab:

Following an introduction to the fundamental types of bonding, students must develop a procedure to characterize an unknown solid as ionic, polar covalent, nonpolar covalent, or metallic. Students then answer questions meant to probe their understanding of the structure-property relationship. (50–90 min.)

Advanced Inquiry Lab:

In the absence of an introductory activity students must develop a procedure to characterize an unknown solid as ionic, polar covalent, nonpolar covalent, or metallic. (60–90 min.)

Outcomes:

Students come to understand that a material's properties are linked to the material's structure and bonding.

Associated Phenomena:

Why are some materials good for certain applications but not others?

Standards

Science & Engineering Practices	Disciplinary Core Ideas	Crosscutting Concepts
Planning and carrying out investigations Analyzing and interpreting data Constructing explanations and designing solutions	HS-PS1.A: Structure and Properties of Matter	Patterns Structure and function Cause and effect

Performance Expectations

HS-PS1-3: Plan and conduct an investigation to gather evidence to compare the structure of substances at the bulk scale to infer the strength of electrical forces between particles.

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**Monash University. Victorian College
of Pharmacy. Department of
Chemistry**

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Government Reports Announcements & Index, 1984 **Introduction to Semimicro Qualitative Analysis** Clarence

Harvey Sorum, J. J. Lagowski, 1977

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supplement for courses with a qualitative analysis component this lab manual contains explanations of the chemistry of metal ions and anions It includes pre lab exercises experiments and lab reports Introduction to Qualitative Analysis Durward C.

Layde, Daryle Hadley Busch, 1968

Qualitative Analysis William A Hardcastle, 2009-11-30 This Guide is concerned with the practice of qualitative analysis and contains invaluable advice for laboratories undertaking such work It is designed to assist laboratories in establishing and maintaining a working regime that will minimise the risk of error and maximise the quality of the analytical information they produce It is aimed primarily at laboratory managers and those responsible for any part of the analysis process from designing qualitative tests or deciding which tests to apply in particular cases to those performing the requisite experimental work Qualitative Analysis has been developed using the knowledge and experience of an advisory panel drawn from a representative cross section of practising analytical scientists It was produced as part of the VAM Programme and will be a useful guide for laboratory staff government agencies researchers and professionals in industry and academia alike **A Brief Introduction to Qualitative Analysis: for Use in Instruction in Chemical**

Laboratories Ludwig Medicus, 1892

QUALITATIVE ANALYSIS AS A LAB William Conger 1874- Morgan, 2016-08-28 This work has been selected by scholars as being culturally important and is part of the knowledge base of civilization as we know it This work was reproduced from the original artifact and remains as true to the original work as possible Therefore you will see the original copyright references library stamps as most of these works have been housed in our most important libraries

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Notes on Qualitative Analysis Monash University. Victorian College of Pharmacy. Department of Chemistry,1955 **Practical Chemistry. The Principles of Qualitative Analysis** Sir William Augustus Tilden,1880 *Tables of Qualitative Analysis* Henry George Madan,1881

Qualitative Chemical Analysis Albert B. Prescott,2015-06-26 Excerpt from Qualitative Chemical Analysis A Guide in Qualitative Work With Data for Analytical Operations and Laboratory Methods in Organic Chemistry Since the last revision of this very comprehensive text on Qualitative Analysis was issued a considerable number of new methods of analysis have been published and some of these have been found to be valuable and have come into more or less general use Many of the new methods have been tried out and some have been found to be unreliable unless extreme care is taken or only pure solutions used The attempt has been made to include in this revision only those methods which have been found to be reliable A very marked change has also taken place in the method of presentation of chemical reactions Ionic reactions are given in many texts to the exclusion of molecular reactions Physical chemical theories are also frequently presented and discussed at great length The attempt has been made in the revision of this text to present only briefly the modern conceptions of solution leaving a fuller presentation for the lecturer or separate texts on Physical Chemistry Molecular reactions have also been largely retained in the belief that the material to be analyzed by the chemist in practical work is quite as often a molecular compound as an ion All molecular weights solubilities and other constants of the elements and their principal compounds have been brought up to date The principal minerals methods of preparation and determinations of elements as well as the reactions have been revised references being given to the literature as heretofore In general the attempt has been made to retain the excellent features of this text which have given it such an extended use in the past both as a class room and as a reference text while adding the valuable results of recent progress in the science About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books Find more at www.forgottenbooks.com This book is a reproduction of an important historical work Forgotten Books uses state of the art technology to digitally reconstruct the work preserving the original format whilst repairing imperfections present in the aged copy In rare cases an imperfection in the original such as a blemish or missing page may be replicated in our edition We do however repair the vast majority of imperfections successfully any imperfections that remain are intentionally left to preserve the state of such historical works

Qualitative Analysis and Chemical Equilibrium Thorfin Rusten Hogness,Warren Charles Johnson,1937 **Manual**

of Qualitative Analysis for Use in the Chemical Laboratory of Dartmouth College Leon Burr Richardson, Andrew Jackson Scarlett, 1921 **The Compendious Manual of Qualitative Chemical Analysis** C. W. Eliot, 2015-06-28 Excerpt from The Compendious Manual of Qualitative Chemical Analysis The authors have endeavored to include in this short treatise enough of the theory and practice of qualitative analysis in the wet way to bring out all the reasoning involved in the subject and to give the student a firm hold upon the general principles and methods of the art It has been their aim to give only so much of mechanical detail as is essential to an exact comprehension of the methods and to success in the actual experiments Hence the multiplication of different tests or processes having essentially the same object has been purposely avoided For the same reason none of the rare elements are alluded to The manual is intended to meet the wants of the general student to whom the study is chiefly valuable as a means of mental discipline and as a compact example of the scientific method of arriving at truth To professional students who wish to make themselves expert analysts this little book offers a logical introduction to the subject an outline which is trustworthy as far as it goes but which needs to be filled in and enlarged by the subsequent use of some more elaborate treatise as a book of reference Professor Johnson of Yale has supplied this need with his excellent edition of Fresenius s comprehensive manual The authors believe that they have put into the following pages as much of inorganic qualitative analysis as is useful for training and also as much as the engineer physician agriculturist or liberally educated man needs to know The book has been written for the use of classes in the Institute of Technology who have already studied the authors Manual of Inorganic Chemistry It is simply an implement devised to facilitate the giving of thorough instruction to large classes in the laboratory It is the authors practice to examine their classes orally every four or five exercises in order to secure close attention to the reasoning of the subject About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books Find more at www.forgottenbooks.com This book is a reproduction of an important historical work Forgotten Books uses state of the art technology to digitally reconstruct the work preserving the original format whilst repairing imperfections present in the aged copy In rare cases an imperfection in the original such as a blemish or missing page may be replicated in our edition We do however repair the vast majority of imperfections successfully any imperfections that remain are intentionally left to preserve the state of such historical works **Practical Chemistry** William A. Tilden, 1907 *Introduction to qualitative analysis* William Lash Miller, 1896

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