STUDY GUIDE FOR CONTENT MASTERY

## The Structure of the Atom

## Section 4.1 Early Theories of Matter

In your textbook, read about the philosophers, John Dalton, and defining the atom.

For each statement below, write true or false.

- false true
- 1. Ancient philosophers regularly performed controlled experiments.
- Philosophers formulated explanations about the nature of matter based on their own experiences.
- true -
- 3. Both Democritus and Dulton suggested that matter is made up of atoms.
- Dalton's atomic theory stated that atoms separate, combine, or rearrange in chemical reactions.
- false
- 5. Dulton's atomic theory stated that matter is mostly empty space.
- Dalton was correct in thinking that atoms could not be divided into smaller particles.
- +rue-
- Dalton's atomic theory stated that atoms of different elements combine in simple whole-number ratios to form compounds.
- false.
- Dalton thought that all atoms of a specific element have the same mass.
  Democritus proposed that atoms are held together by chemical bonds, but no one believed him.
- true.
- Dulton's atomic theory was based on careful measurements and extensive research.
- -false
- There are no instruments powerful enough to magnify atoms so that they can be seen.
- true
- The smallest particle of an element that retains the properties of that element is called an atom.

# **Quantum Theory The Atom Study Guide Answers**

**Clyde Metz** 

### **Quantum Theory The Atom Study Guide Answers:**

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