

REPORT SHEET

Rates of Chemical Reaction

Name _____

Date _____

Section _____

Partner's Name _____

Instructor _____

(1.5 pts. total for Tables 1-3)

Table 1, Experiment #1

Initial moles of $S_2O_8^{2-}$		Cumulative time recorded from stopwatch		Cumulative time converted to seconds (s)	Mole of $S_2O_8^{2-}$ Consumed	Mole of $S_2O_8^{2-}$ Remaining	Concentration of $S_2O_8^{2-}$ Remaining (mol/L)
Aliquot #		(min)	(sec)				
1		1 min	46 sec	106	$1.0 \times 10^{-4} \text{ mol}$	0.025 mol	0.046 M
2		3	24 sec	204	$2.0 \times 10^{-4} \text{ mol}$	0.023 mol	0.043 M
3		4	29 sec	269	$3.0 \times 10^{-4} \text{ mol}$	0.021 mol	0.041 M
4		5	48 sec	345	$4.0 \times 10^{-4} \text{ mol}$	0.020 mol	0.038 M
5		7	03	423	$5.0 \times 10^{-4} \text{ mol}$	0.019 mol	0.035 M

Table 2, Experiment #2

Initial moles of $S_2O_8^{2-}$		Cumulative time recorded from stopwatch		Cumulative time converted to seconds (s)	Mole of $S_2O_8^{2-}$ Consumed	Mole of $S_2O_8^{2-}$ Remaining	Concentration of $S_2O_8^{2-}$ Remaining (mol/L)
Aliquot #		(min)	(sec)				
1		2	45	165	$1.0 \times 10^{-4} \text{ mol}$	0.023 mol	0.046 M
2		5	19	319	$2.0 \times 10^{-4} \text{ mol}$	0.022 mol	0.043 M
3		8	33	513	$3.0 \times 10^{-4} \text{ mol}$	0.021 mol	0.041 M
4		11	11	671	$4.0 \times 10^{-4} \text{ mol}$	0.020 mol	0.038 M
5		13	17	797	$5.0 \times 10^{-4} \text{ mol}$	0.019 mol	0.035 M

Table 3, Experiment #3

Initial moles of $S_2O_8^{2-}$		Cumulative time recorded from stopwatch		Cumulative time converted to seconds (s)	Mole of $S_2O_8^{2-}$ Consumed	Mole of $S_2O_8^{2-}$ Remaining	Concentration of $S_2O_8^{2-}$ Remaining (mol/L)
Aliquot #		(min)	(sec)				
1		1	20	80	$1.0 \times 10^{-4} \text{ mol}$	0.024 mol	0.034 M
2		2	30	150	$2.0 \times 10^{-4} \text{ mol}$	0.024 mol	0.035 M
3		3	43	223	$3.0 \times 10^{-4} \text{ mol}$	0.025 mol	0.037 M
4		3	59	233	$4.0 \times 10^{-4} \text{ mol}$	0.024 mol	0.033 M
5		5	06	306	$5.0 \times 10^{-4} \text{ mol}$	0.023 mol	0.030 M

Rates Of Chemical Reactions Lab

Jianjun Gao



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