

A crossword puzzle grid is shown. The grid consists of white squares for letters and black squares for empty space. The grid is 15 rows by 15 columns. The black squares are located at the following coordinates (row, column): (1, 10), (2, 10), (3, 10), (4, 10), (5, 10), (6, 10), (7, 10), (8, 10), (9, 10), (10, 13), (11, 10), (12, 10), (13, 10), (14, 10), (15, 10).

2. The area of chemistry involving the calculations of quantities of substances involved in chemical reactions
3. A representation of a chemical reaction using symbols and numbers to show the relationships between the reactants and products
4. $AB \rightarrow A + B$
6. The percent by mass of each element in a compound
10. The ratio of the actual yield as compared to the theoretical yield expressed as a percentage
12. A conversion factor derived from the coefficients of the quantities of substances involved in chemical reactions

- molar ratio
- percent composition
- percent yield
- actual yield
- reactants

synthesis
double replacement
chemical equation
limiting reactant

13. 6.02×10^{23}
15. The chemicals which are put into a chemical reaction and are on the left side of the reaction
18. $A + B \rightarrow AB$
19. $AB + CD \rightarrow AD + CB$
20. The substance that runs out in a chemical reaction, thus controlling the amount of product(s)
21. The amount of product that could form based on a balanced chemical equation
Down
1. A compound containing carbon and hydrogen is burned in the presence of oxygen
5. $A + BC \rightarrow AC + B$

stoichiometry
avogadro's number
theoretical yield
excess

7. The mass of one mole of a substance
8. The measurement of the mass of the substances in a chemical reaction
9. The chemicals present on the right side of a chemical reaction, that are only present after the the chemical reaction has begun
11. The reactant that is leftover after a reaction comes to completion
14. The amount of a substance
16. The amount of product that is actually formed when a reaction is carried out in the laboratory
17. The number in front of a balanced formula showing how much of that reactant or product is present

molarmass:
coefficient:
combustion:
gram

decomposition
single replacement
mole
products

Section Chemical Formulas And Equations Crossword Answers

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one of my students At the time that question compelled me to answer in a way I didn't like namely I'll think about it and I hope I'll have the answer by the next time we meet I didn't have it that soon though I did manage to have it before the end of the course

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