

HONORS CHEMISTRY: FINAL EXAM STUDY GUIDE

- The final exam will consist of 80 Multiple Choice questions and 2 open-end questions. The outline provided below describes the major topics and units that will be covered on the multiple choice section of the final exam:

UNIT 6: "Gases and Gas Law Relationships" (~ 10 questions)

- Gases vs. Liquids vs. Solids: Properties and Behavior
- Pressure and Volume: Boyle's Law
- Temperature and Volume: Charles' Law
- The Combined Gas Law
- Moles and The Ideal Gas Law Equation
- Avogadro's Law, Dalton's Law of Partial Pressure, Graham's Law

UNIT 7: "Chemical Bonding and Molecular Geometry" (~ 20 questions)

- Ionic, Covalent and Metallic Bonding: Definitions, Models and Properties
- Using Dot Diagrams to draw Lewis bond structures
- Polar vs. Non-Polar Bonding and Electronegativity Difference
- Isomers and Resonance
- VSEPR Theory, Facts, and Examples
- AB_n Formulas, Bond Length and Bond Angles, Molecule Shapes
- Bond Hybridization Basics
- Intermolecular Forces including Hydrogen Bonding

UNIT 8: "Thermodynamics and Kinetics" (~ 20 questions)

- Definitions of Energy, Enthalpy, Entropy
- Exothermic vs. Endothermic Changes
- Hess's Law Calculations and Interpretation
- Entropy Analysis and Factors Affecting Entropy Change
- Gibbs Free Energy Calculations and Interpretation
- Rate-Influencing Factors and Collision Theory
- Determining Rate Order and Solving Rate Law Expressions
- Reaction Mechanisms
- Catalysts and Energy/Rate Graphical Analysis

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United States Air Force Academy



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Anne came from a well to do family that had some naval and theological importance in England Morris born on January 24th 1872 and his siblings were cared for by a series of nannies nurses and maids Though William s medical practice was doing well his children lived simply Pocket money for the children was limited though William and Anne did take their family on many happy country holidays Morris received his early education at Ramsgate Woking College and at Blundell s School While at Blundell s he excelled in Chemistry but had significant difficulty with Latin He entered University College in London in 1891 It is likely that an interest in science for Morris Travers had a good deal to do with his father who always felt he had to read more about his field just to keep up and visits to the Regent Street Polytechnic and the Electrical Exhibition at the Crystal Palace He unlike most of his classmates enjoyed the weekly chemistry lectures he received at Blundell s William s encouragement was also important in Morris s choosing to study at University College in London Though he enjoyed zoology Morris soon fell under the tutelage of Ramsay and Plimpton He earned his B Sc in 1893 and began advanced study of chemistry later that same year Though he had entertained working with Haller in France to earn his Ph D Morris became fascinated with the discovery of a new gas by Ramsay helium and began work on the properties of the new gas He was invited by Ramsay to stay and look for the element with mass of 20 This effort led to the discovery of neon Further work with Ramsay and Hampson on liquefying the components of air brought about the discovery of first krypton and then xenon Morris who had great mechanical aptitude and some ability as a draftsman then was given the task of isolating hydrogen to separate neon from a mixture of air and helium He built an apparatus that so allowed the isolation of liquid and solid hydrogen to make such a separation for neon and then began a study of its properties He is first asked if he would be interested in being the Director of the proposed Indian institute of Science and publishes his first book The Experimental Study of Gases He works on developing temperature scales by measuring the vapor pressures of liquid hydrogen and liquid oxygen meets his future wife Dorothy Gray and begins searching for an academi *University of Colorado at Denver Catalog* University of Colorado at Denver,2002 **Cornell University Description of Courses** Cornell University,1976

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