

Section 6.2 Covalent Bonding

(pages 165–168)

This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules.

Reading Strategy (page 165)

Relating Text and Models As you read the section, look closely at Figure 9. Complete the table by describing each type of model shown. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Molecular Models	
Model	Description
Electron dot	
Structural formula	
Space-filling	
Electron cloud	

Covalent Bonds (pages 165–167)

- Describe a covalent bond.
- Circle the letters of molecular models that show overlap of atoms overlapping when a covalent bond forms.
a. electron dot b. structural formula
c. space-filling d. electron cloud
- Describe a molecule.
- Is the following sentence true or false?
"In a covalent bond, the atoms are held together by the attractions between the shared electrons and the protons in each nucleus."
- Circle the correct answer. Nitrogen has five valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
a. zero b. one
c. two d. three

Section 6 2 Covalent Bonding Answer Key

Ensheng Dong



Section 6 2 Covalent Bonding Answer Key:

Section 6 2 Covalent Bonding Answer Key Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the energy of words has are more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such could be the essence of the book **Section 6 2 Covalent Bonding Answer Key**, a literary masterpiece that delves deep into the significance of words and their affect our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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