

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride SF_6
- b. Lithium nitride Li_3N
- c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride SnCl_2
- e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)
- f. Mercury (I) chloride Hg_2Cl_2
- g. Potassium bromate KBrO_3
- h. Hydrosulfuric acid $\text{H}_2\text{S(aq)}$
- i. Chloric acid HClO_3
- j. Sulfurous acid H_2SO_3

Name each of the following compounds.

- a. CuSO_4 Copper (II) sulfate
- b. AlF_3 aluminum fluoride
- c. HI hydrogen iodide
- d. NO nitrogen monoxide
- e. $\text{H}_2\text{Se(aq)}$ hydroselenic acid
- f. HNO_3 nitric acid
- g. HNO_2 nitrous acid
- h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

61.5 yr

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} \boxed{12.5 \text{ mg}}$$

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 \times .692 & = & 43.596 \rightarrow 44 \\ 65 \times .308 & = & 20.02 \rightarrow 20. \\ \hline & + & \\ & 63.62 & \rightarrow 64 \end{array}$$

Section 3 1 Chemistry Answers

Chao Zhang



Section 3 1 Chemistry Answers:

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TABLE OF CONTENT

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 2019 IIT JEE 2006 IIT JEE 2007 IIT JEE 2008 IIT JEE 2009 IIT JEE 2010 IIT JEE 2011 IIT JEE 2012 JEE Main Advanced 2013
 JEE Main Advanced 2014 JEE Main Advanced 2015 JEE Main Advanced 2016 JEE Main Advanced 2017 JEE Main Advanced
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 prescribed pattern and types of questions With this helpful set of Chapterwise solved papers students will be ensured to get
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TABLE OF CONTENT

Some Basic Principles of Chemistry Atomic Structure Chemical Bonding
 Solutions States of Matter Nuclear Chemistry Chemical Equilibrium Ionic Equilibrium Thermodynamics Chemical Kinetics
 Electrochemistry Surface Chemistry Metallurgical Operations Chemical Periodicity Hydrogen and its Compounds and s Block
 Elements p Block Elements Transition Elements d and f Block Elements Coordination Compounds Chemical Analysis General
 Organic Chemistry Hydrocarbons Alkyl Halides Alcohols Phenols and Ethers Aldehydes And Ketones Carboxylic Acids and
 their Derivatives Organic Compounds Containing Nitrogen Polymers Biomolecules and Chemistry in Everyday Life Appendix
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Table of Contents Section 3 1 Chemistry Answers

1. Understanding the eBook Section 3 1 Chemistry Answers
 - The Rise of Digital Reading Section 3 1 Chemistry Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Section 3 1 Chemistry Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 3 1 Chemistry Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section 3 1 Chemistry Answers
 - Personalized Recommendations
 - Section 3 1 Chemistry Answers User Reviews and Ratings
 - Section 3 1 Chemistry Answers and Bestseller Lists
5. Accessing Section 3 1 Chemistry Answers Free and Paid eBooks

- Section 3 1 Chemistry Answers Public Domain eBooks
- Section 3 1 Chemistry Answers eBook Subscription Services
- Section 3 1 Chemistry Answers Budget-Friendly Options
- 6. Navigating Section 3 1 Chemistry Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Section 3 1 Chemistry Answers Compatibility with Devices
 - Section 3 1 Chemistry Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section 3 1 Chemistry Answers
 - Highlighting and Note-Taking Section 3 1 Chemistry Answers
 - Interactive Elements Section 3 1 Chemistry Answers
- 8. Staying Engaged with Section 3 1 Chemistry Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 3 1 Chemistry Answers
- 9. Balancing eBooks and Physical Books Section 3 1 Chemistry Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 3 1 Chemistry Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Section 3 1 Chemistry Answers
 - Setting Reading Goals Section 3 1 Chemistry Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Section 3 1 Chemistry Answers
 - Fact-Checking eBook Content of Section 3 1 Chemistry Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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