Energy Changes in Chemical reactions for KS3 Science - Worksheet

- In an exothermic reaction does the temperature go up or down?
- In an endothermic reaction does the temperature go up or down?
- 3. Name two examples of exothermic reactions
- 4. Name two examples of endothermic reactions
- Circle the correct answers.

The bonds between the atoms of the reactants / products need to be broken first, this is an endothermic / exothermic process. Theme bonds are made between the atoms of the reactants / products, this is an endothermic / exothermic process.

6. Use the table to answer this question

Reaction	Starting temperature °C	Final temperature °C
A	20	31
В	22	18
С	21	25

Decide whether each reaction is endothermic or exothermic, explain how you could tell.

b. Which reaction has the largest energy change?



Energy Changes in Chemical reactions for KS3 Science

<u>Section 3 Reinforcement Chemical Reactions And</u> <u>Energy Worksheet Answers</u>

Prentice-Hall Staff

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