

Chapter 5 The Periodic Table

Section 5.2 The Modern Periodic Table

(pages 130–138)

This section explains the organization of the modern periodic table and discusses the general properties of metals, nonmetals, and metalloids.

Reading Strategy (page 130)

Previewing Before you read, complete the table by writing two questions about the periodic table on pages 132–133. As you read, write answers to your questions. For more information on this reading strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Questions About the Periodic Table	
Question	Answer
Students might ask: What does atomic mass mean?	Atomic mass is a value that depends on the distribution of an element's isotopes in nature and the masses of those isotopes.
Why are two columns of elements placed before the main body of the table?	Two kinds of elements are placed before the table to make the table more compact.

Other possible questions: Why are there two numbering systems for the columns? Why is period 7 incomplete?

The Periodic Law (pages 131–133)

1. Is the following sentence true or false? In the modern periodic table, elements are arranged by increasing number of protons.
_____ **true** _____
2. Explain why the number of elements per period varies. _____ **The number of elements per period varies because the number of available orbitals increases from energy level to energy level.** _____
3. Properties of elements repeat in a predictable way when atomic numbers are used to arrange elements into groups. This pattern of repeating properties is called the _____ **periodic law** _____.

Atomic Mass (page 134)

4. Label the four types of information supplied for chlorine in the diagram.

a.	17
b.	Cl
c.	Chlorine
d.	35.453

- Atomic number**
- Element symbol**
- Element name**
- Atomic mass**

Section 5 2 The Modern Periodic Table

Randy Shih



Section 5 2 The Modern Periodic Table:

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into gold to modern chemists synthesizing new elements in the laboratory We ll then delve into the properties of elements examining their physical and chemical characteristics We ll explore how elements react with each other forming compounds with a wide range of properties We ll also discuss the role of elements in our everyday lives from the air we breathe to the food we eat to the medicines we take But elements are not without their challenges We ll examine the dangers posed by radioactive elements the toxicity of some elements and the environmental impact of element mining and processing We ll also discuss the geopolitical issues surrounding access to rare elements and the need for responsible element stewardship Finally we ll look to the future of elements We ll explore the search for new elements the potential of elements for new materials and technologies and the role of elements in space exploration We ll also discuss the impact of elements on the environment and the need for sustainable element use Throughout this journey we ll be amazed by the beauty and diversity of elements We ll discover the incredible power of these tiny particles and the profound impact they have on our world If you like this book write a review

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