

Copper In Silver Nitrate Lab

by Charles Deaton

Background

The purpose of this lab is to find the theoretical percentage yield of the reaction. The reaction consisted of copper(Cu) and silver nitrate(AgNO_3) which after 3 class days of reacting, turned into copper nitrate($\text{Cu}(\text{NO}_3)_2$) and silver(Ag). In that third class day, they massed out that it had copper wire, and Ag and all silver nitrate solution was evaporated. Once it was evaporated, they were trying to see the color of a small beaker with a good portion of the wire submerged in AgNO_3 and of solution. Observations were made and then the beakers were left out until day 3. When students returned a couple days later, and recorded what they noticed in a beaker. Striking them back the wire out of the solution and added with water so that the precipitate were like the beakers. This mass of the wire was taken again, and a piece of filter paper was also weighed. The filter paper was then set up in a dry beaker with a funnel and a large beaker beside. The concentrated HNO_3 to beaker was poured through the filter paper, and mixed with more water. This was to remove all of the reactants from the beaker. Day 4 was to record the final mass of the products so that the paper that could be dry from the days prior. After everything was massed and written in a table, calculations were done to find the theoretical yield, percent yield, and percent error of the lab.

Silver Nitrate Lab Report

**Chemical Laboratory (American
Medical Association)**



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Publications, Reports and Papers for ... from Oak Ridge National Laboratory Oak Ridge National Laboratory, 1964

Water-resources Investigations Report, 1998

Practical Chemistry Labs Leonard Saland, 1989 Features self contained step by step activities using common materials and covering topics from food chemistry to papermaking and electrochemistry Illustrates the connection between the real world and chemistry concepts such as solutions chemistry acids and bases and more Includes teacher notes quizzes and answers to help monitor student progress

Publications, Reports, and Papers for 1961- from Oak Ridge National Laboratory Oak Ridge National Laboratory, 1965

Report summaries United States. Environmental Protection Agency, 1983

Scientific and Technical Aerospace Reports, 1991

Keywords Index to U.S. Government Technical Reports

United States. Department of Commerce. Office of Technical Services, 1963

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Proceedings of Technical Meetings U.S. Atomic Energy Commission, 1961

Bulletin of the Hygienic Laboratory

, 1920

Environmental Sampling and Analysis Maria Csuros, 1997-03-24 This manual covers the latest laboratory techniques state of the art instrumentation laboratory safety and quality assurance and quality control requirements In addition to complete coverage of laboratory techniques it also provides an introduction to the inorganic nonmetallic constituents in environmental samples their chemistry and their control by regulations and standards Environmental Sampling and Analysis Laboratory Manual is perfect for college and graduate students learning laboratory practices as well as consultants and regulators who make evaluations and quality control decisions Anyone performing laboratory procedures in an environmental lab will appreciate this unique and valuable text

Technical Abstract Bulletin, 1965

Biennial

Report Minnesota. Dairy and Food Dept, 1898

Nuclear Science Abstracts, 1976-06

Alcamo's Laboratory

Fundamentals of Microbiology Jeffrey C. Pommerville, 2010 This Popular Lab Manual Offers Thirty Four Multi Part Lab Exercises Designed To Provide Students With Basic Training In The Handling Of Microorganisms While Exploring Microbial Properties And Uses This Lab Manual Can Also Be Used Independently Of The Main Text An Instructor S Manual Downloadable From The Web Accompanies The Lab Manual And Provides Principles Of Lab Safety Research Topic Ideas Information On Customizing Laboratory Programs With The Manual Helpful Suggestions For Setting Up And Running Each Exercise And Lists Of Laboratory Media Cultures And Special Materials Used In Each Exercise

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