Catalytic Hydrogenation	Parc meso *Pt can also be used*	
Ozonolysis (Reducing Conditions)	1. O ₃ 2. (CH ₂) ₂ S 2. Znith ₂ O	
Ozonolysis (Oxidizing Conditions)/Oxidative Cleavage	1. O ₁ 2. H ₂ O ₂ or OH 1. KM+O ₂ /hoost	

Alkyne Reactions

Catalytic Hydrogenation (Catalytic Reduction)	= H ₀
Reduction to Cis- Alkene	— H ₂
Reduction to Trans- Alkene	Na or Li NH ₃ (0
Hydrohalogenation with HBr (Terminal Alkyne)	— 1 eq HBr Br 1 eq HBr Br Br
Hydrohalogenation with HBr (Internal Alkyne)	Ted High St
Halogenation with Br ₂	— 1 eq Br ₂ Br ₃ Br ₄ Br ₅ Br ₄ Br ₄ Br ₅ Br ₄ Br ₅ Br ₅ Br ₅ Br ₆ Br ₅ Br ₆ Br ₆ Br ₇
Hydration of an Internal Alkyne	- Huston - 2 · · ·
Hydration of a Terminal Alkyne (Markovnikov)	■ H ₃ SO ₄ P
Hydration of a Terminal Alkyne (Anti-Markovníkov)	= 1. Sia_BH-THF H
S _{v2} Addition of an Acetylide Ion to an Alkyl Halide	
5 ₆ 2 Addition of an Acetylide Ion to a Ketone	$=$ $\xrightarrow{\text{NaNH}_2}$ $=$ $\stackrel{\text{P}}{=}$ $\stackrel{\text{L.}}{\stackrel{\text{L.}}{\downarrow}}$ $\stackrel{\text{O}}{=}$ $\stackrel{\text{O}}{=}$ $\stackrel{\text{O}}{=}$ OH
S _v 2 Addition of an Acetylide Ion to an	

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Solution Manualessential Organic Chem

Thomas Griffiths

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